



Anticancer Drugs

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Introduction

- The diversity of tumors and the similarity to normal cells are the main obstacle preventing the reach of an ultimate remedy.
- The main task of anticancer drug is to **destruct** malignant cells, while **sparing** normal cells.
- Cancers could be classified according to their nature and location throughout the human body into two main categories: **Solid tumors** such as glandular tissue cancers, connective tissue cancers. **Malignant hematologic diseases** such as lymphatic ganglia and blood cancers.
- Tumors can be classified according to their locations into:
 - Carcinoma → Glandular tissue cancers.
 - Sarcoma → Connective tissue cancers.
 - Lymphoma → Lymphatic ganglia cancers.
 - Leukemia → Blood cancers.

- Solid tumors are difficult to treat due to the **lack of vascularity** and **blood** supply inside the tumor, which prevents any drug from reaching the core of the tumor mass.
- The rate of cell division is the only sensible difference (found so far) which distinct cancer cell from normal cell, tumor cell is rapidly proliferating.
- The lack of obvious difference makes it difficult for any chemotherapeutic agent to distinguish between cancer cell and any healthy cell, especially those which are naturally of rapid cell division, as for example hair, **bone marrow cells** and the **mucosa lining the walls** of the gastrointestinal tract.

How could the anticancer agents be selectively toxic to cancer cell?

- Tumor cell is more rapidly proliferating than normal cell so it will consume more of the drug. But the drug is highly toxic to the organs which is normally rapidly proliferating such as bone marrow, hair, GIT.

- Anticancer agents could be classified based on type of activity into three main groups:

1) **Growth inhibitors:** Drugs which inhibit the growth of cancer cells into 50%.

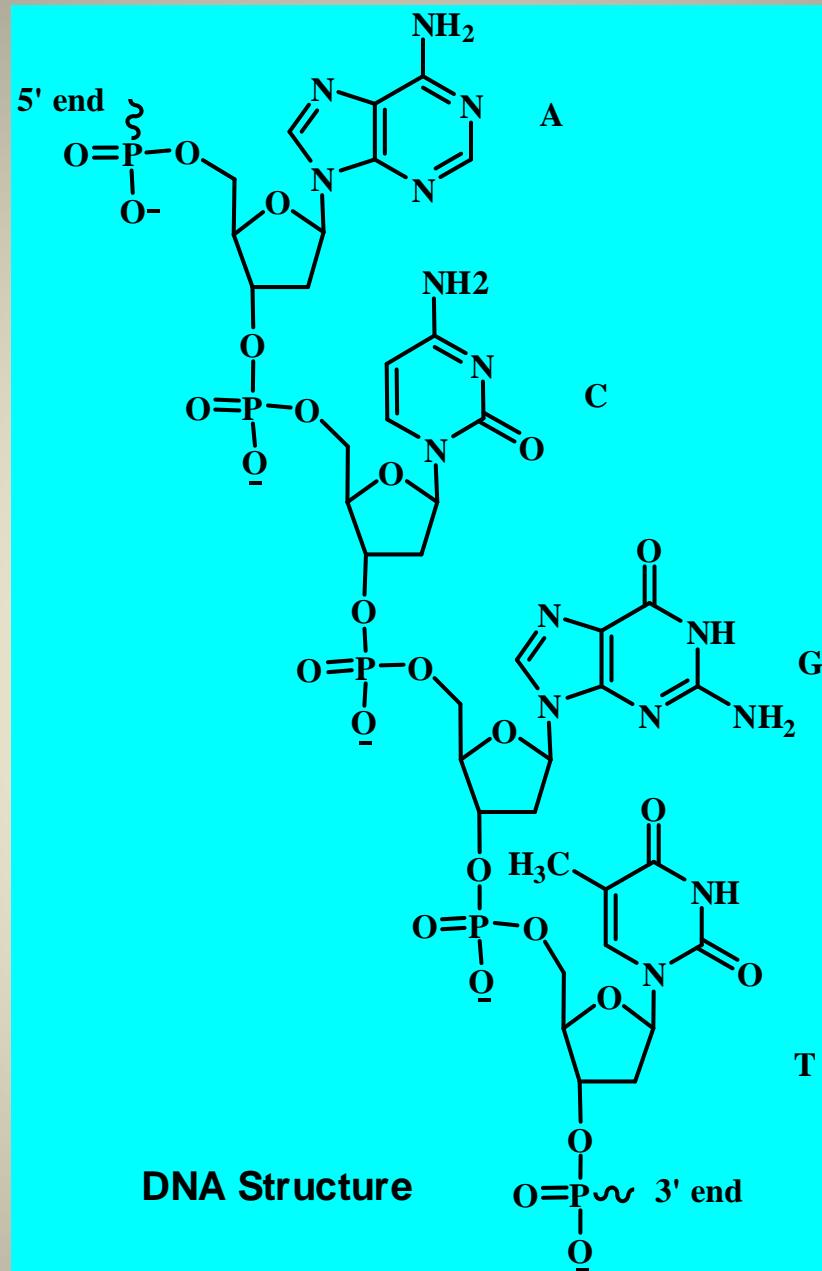
Median growth inhibitory concentration (GI_{50}): Molar concentration which inhibit net cell growth to 50%.

2) **Cytostatic agents:** Drugs which totally inhibit the growth of cancer cells.

Total growth inhibitory concentration (TGI): Molar concentration which cause total inhibition of cell growth.

3) **Cytotoxic agents:** Drugs which cause 50% killing of the original no. of cancer cells.

Median lethal concentration (LC_{50}): Molar concentration which cause 50% killing of the initial cell level



Cancer Chemotherapy

- The difference between normal and cancerous cell lies in the cell nucleus which controls cell division and rather more, it might reach the gene level.
- Most of the antineoplastic agents are designed to interfere with the protein synthesis followed by the inhibition of cell vital processes leading to cell death.
- Anticancer agents could be classified based on their mode of action into:
 - I. DNA Interactive Drugs (DID)
 - II. Antimetabolites
 - III. Hormones

I. DNA Interactive Drugs (DID)

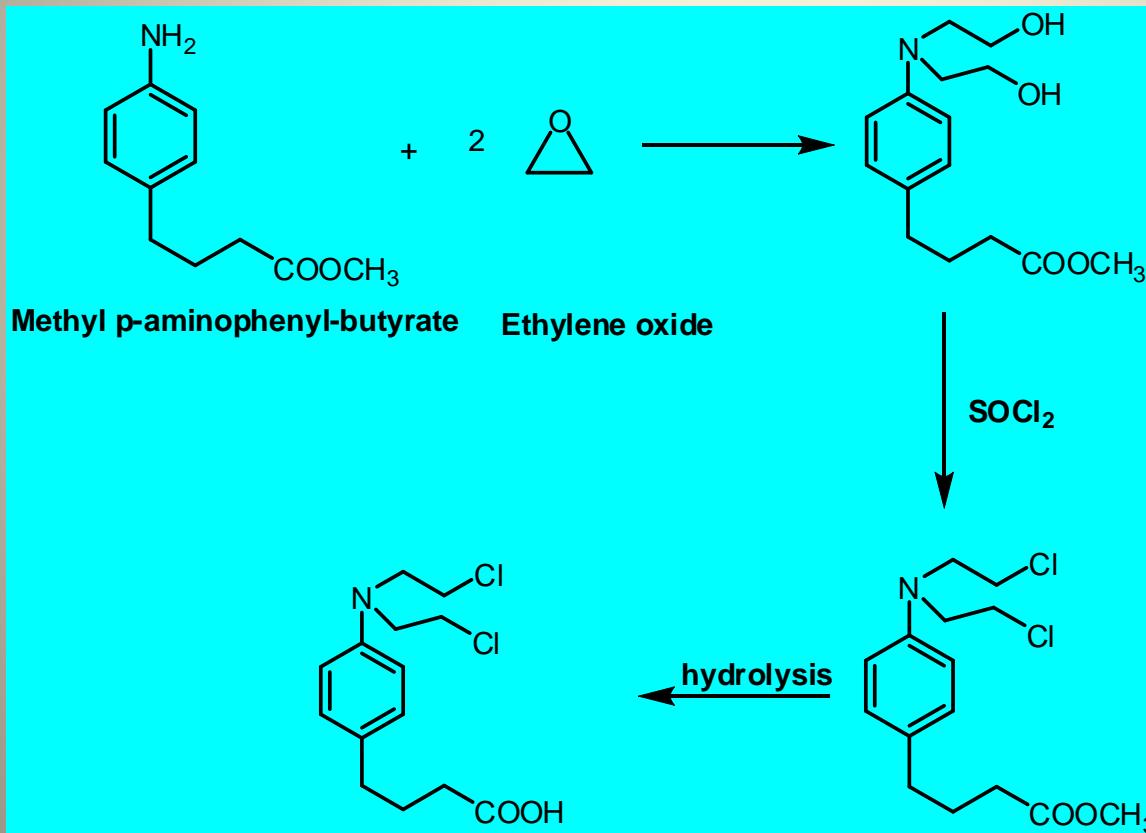
A. DNA Alkylators

- **DNA alkylators are those class of compounds which proved to alkylate the nucleophilic centers at the nucleic acid bases (guanine, thymine ... etc.) leading to the formation of **deformed DNA** which will affect protein synthesis causing **cell death**.**
- **DNA alkylators could be classified into:**
 - a. Nitrogen mustard and its analogues.
 - b. Ethylenimines.
 - c. Epoxides.
 - d. Sulphonic acid esters.
 - e. Miscellaneous.

Nitrogen Mustards

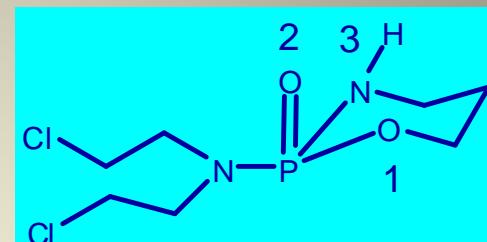
1. Chlorambucil, Leukeran

4-[Bis-(2-chloroethyl)amine]phenylbutyric acid.

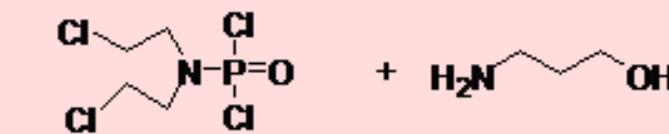


2. Cyclophosphamide, Cytoxin, Endoxan

N,N-Bis(2-chloroethyl)tetrahydro-2*H*-1,3,2-oxazaphosphorin-2-amine-2-oxide.

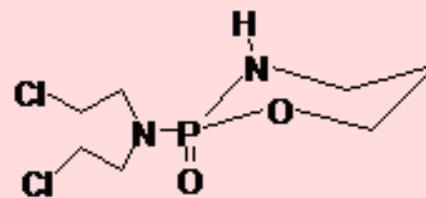


Synthesis

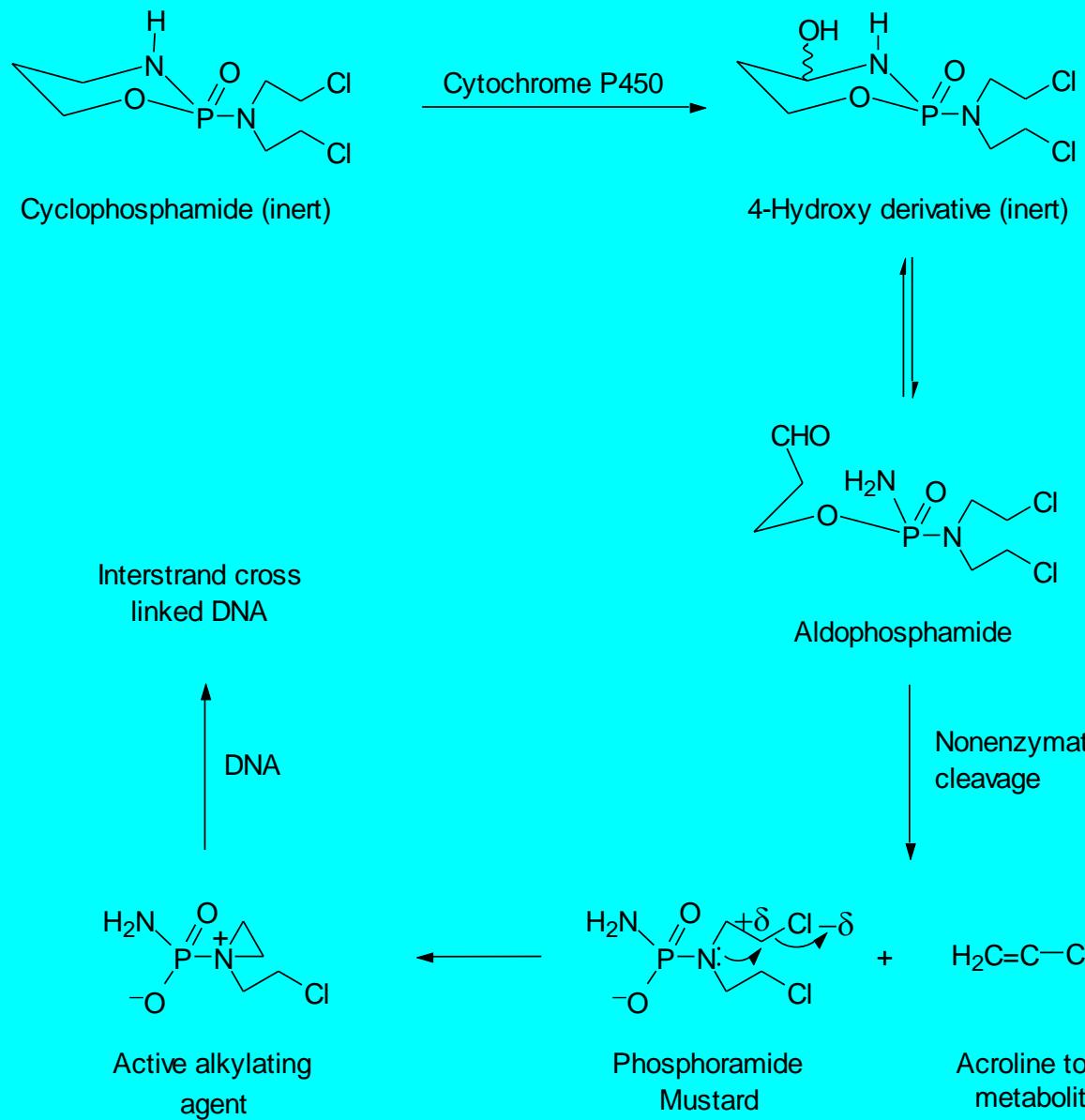


Bis(2-chloroethyl)phosphoramide dichloride

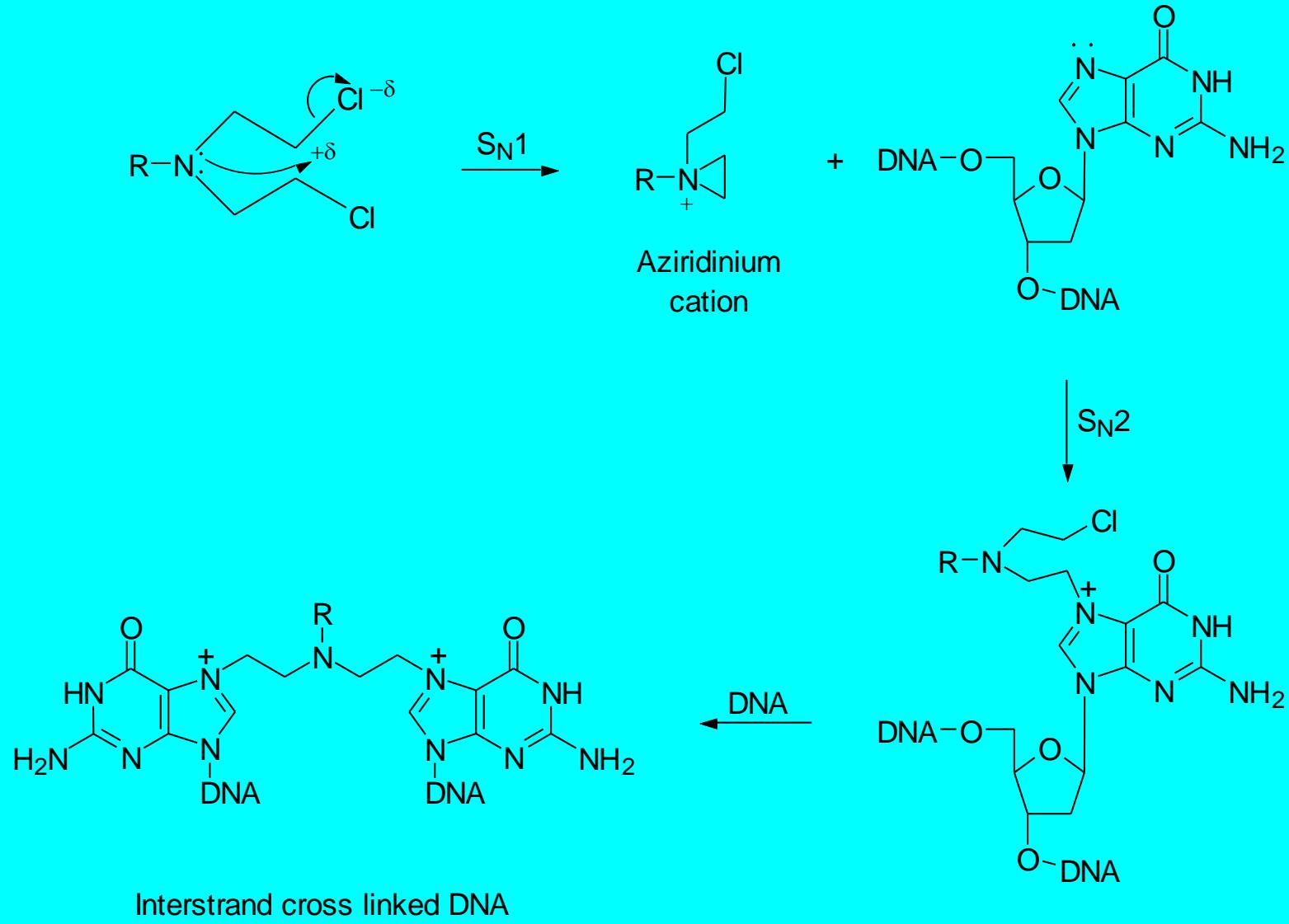
Propanolamine



**N,N-bis(2-chloroethyl)tetrahydro-2*H*-1,3,2-oxazaphosphorin-2-amine-2-oxide,
(cyclophosphamide).**



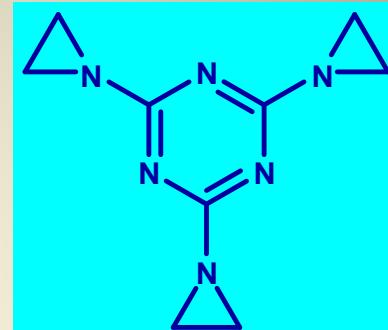
Mode of Action of Nitrogen Mustards



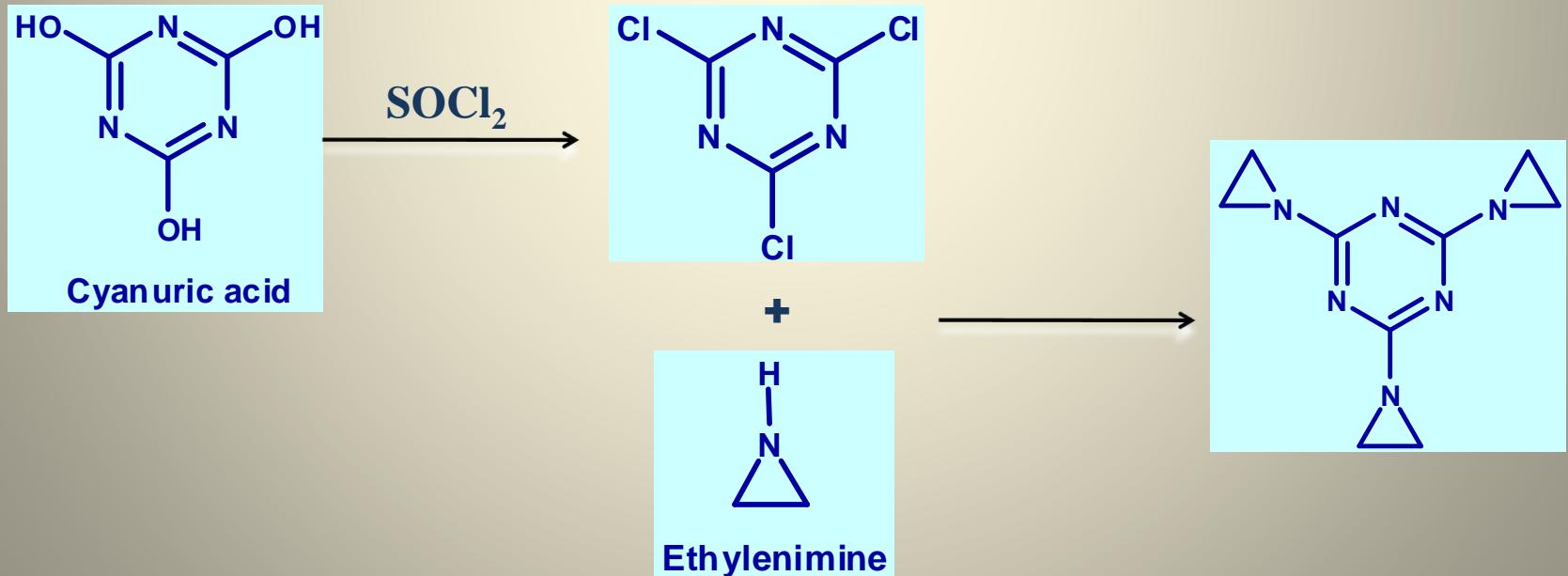
Ethylenimine Derivatives

1. Triethylene melamine (TEM)

2,4,6-Tris (1-aziridinyl)-s-triazine.

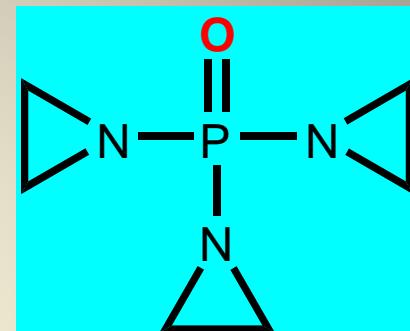


Synthesis

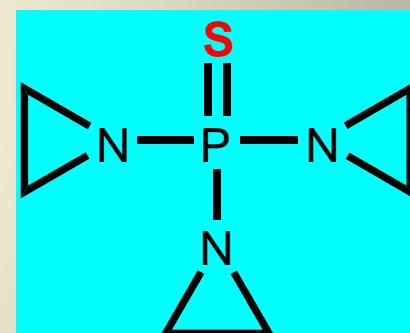


2. Phosphoramido Derivatives

Triethylene phosphoramide (TEPA)

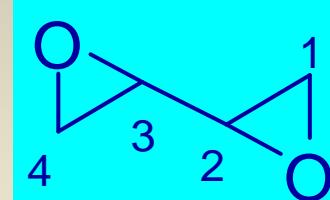


Triethylene thiophosphoramide (thio-TEPA)

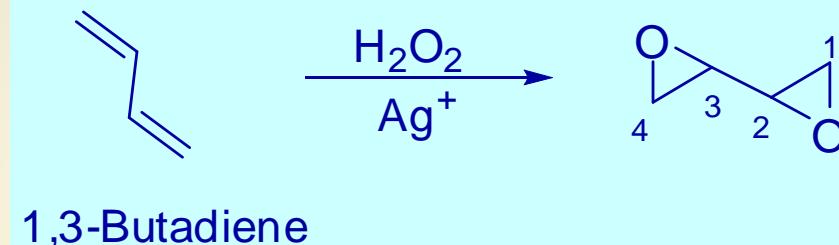


Epoxides

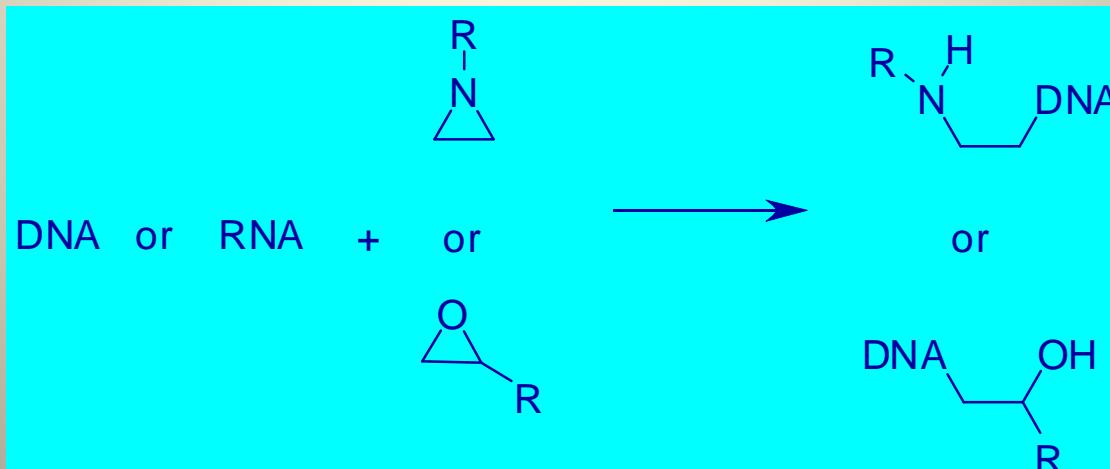
1,2,3,4-Diepoxybutane



Synthesis



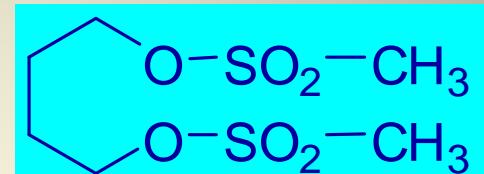
Mode of Action of Ethylenimines and Epoxides



Sulphonic Acid Esters

Busulfan, Myleran

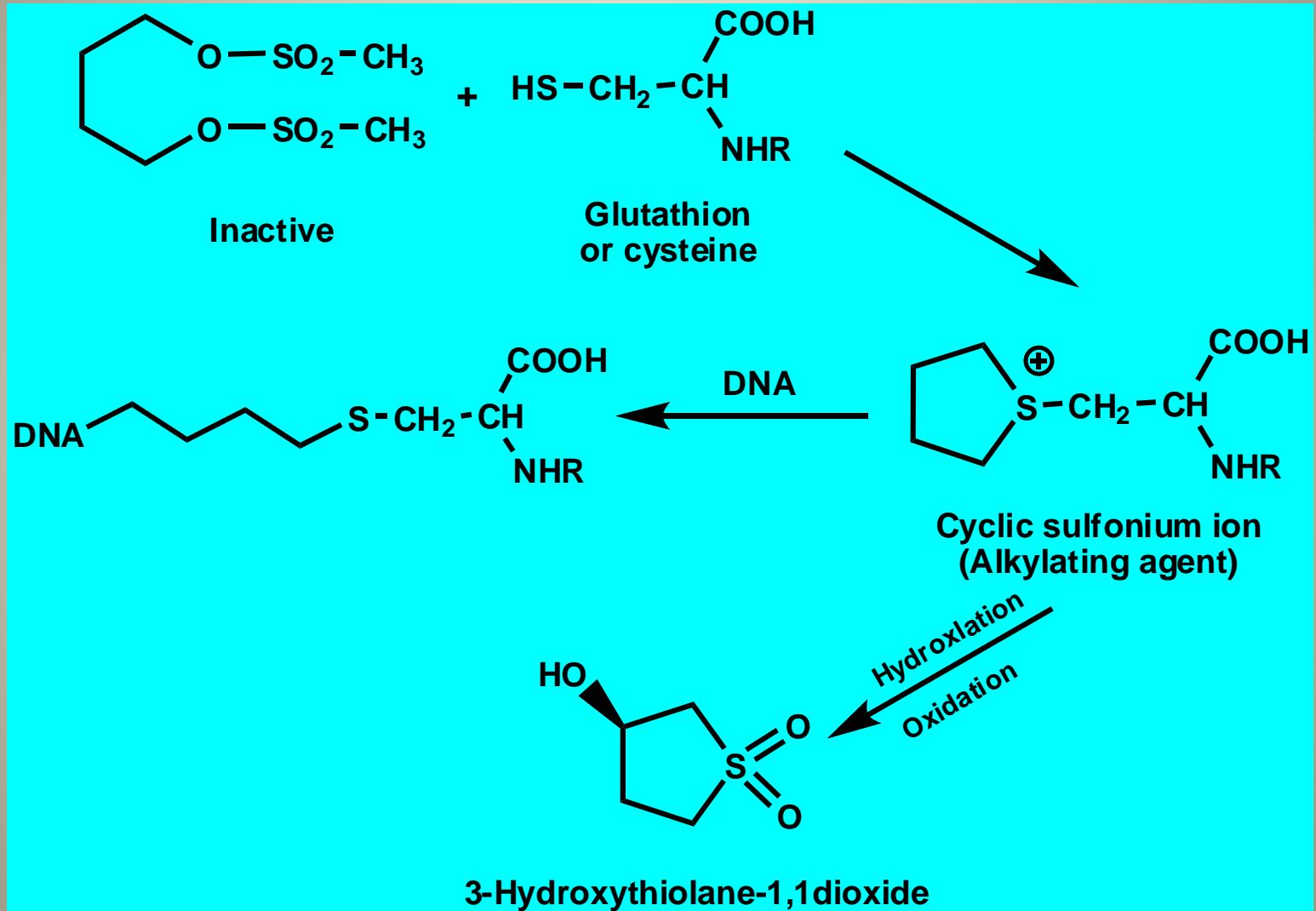
1,4-Di(Methan-sulphonyl-oxy)butane



Synthesis



Metabolism and Mode of Action



Miscellaneous

1. Dacarbazine, DTIC

5-(3,3-Dimethyl-1-triazino)-imidazole-4-carboxamide



2. Procarbazine HCl, Matulane

N-(Isopropyl)-4-(2-methyl-hydrazino-methyl)benzamide



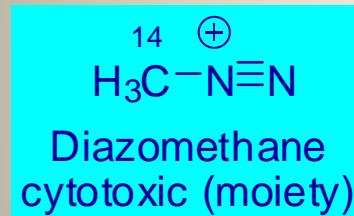
Metabolism and Mode of Action



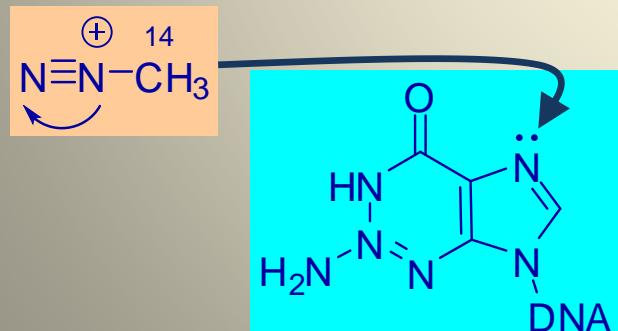
Oxidative
dealkylation
Cytochrome
P-450



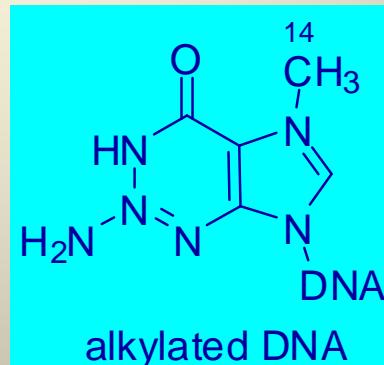
Monomethyl
derivative



+



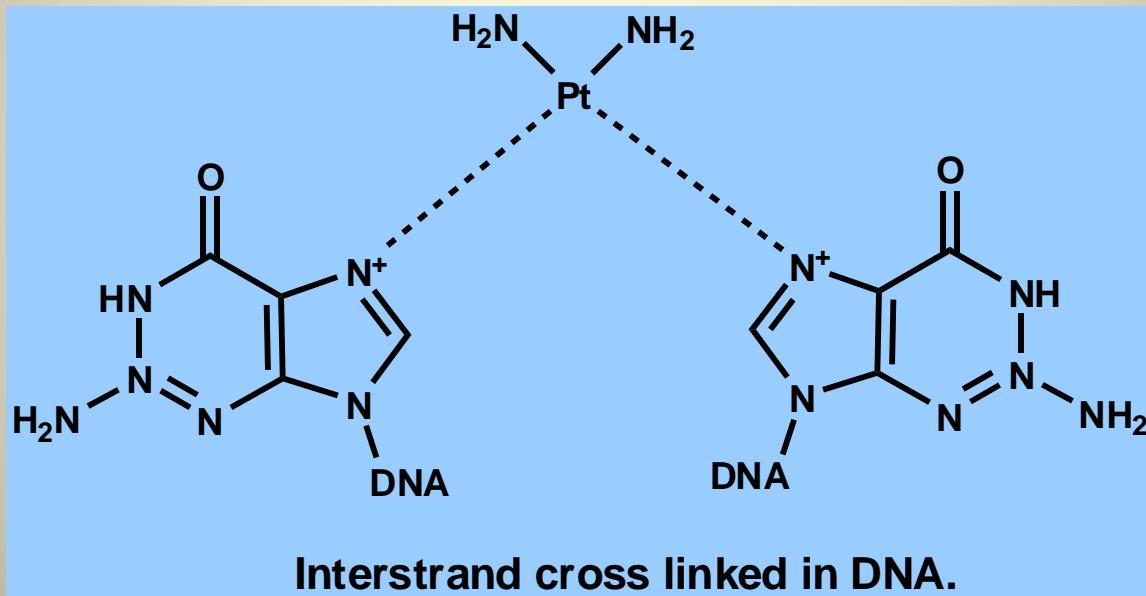
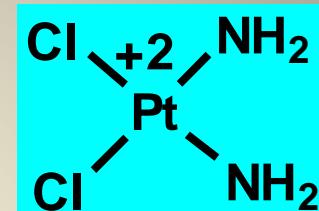
$-\text{N}_2$



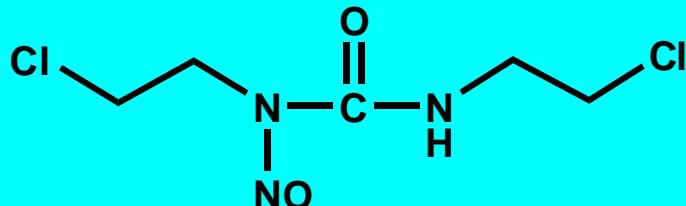
**Cell
Death**

3. Platinum Compounds

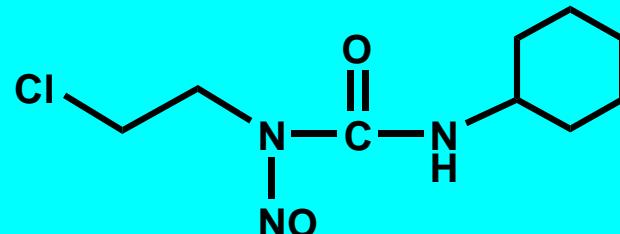
Cisplatin, Platinol,
Diaminedichloroplatinum



4. Nitrosoureas



Carmustine (BCNU)
1,3-Bis(2-chloroethyl)-1-nitrosourea



Lumustine (CCNU)
1-(2-chloroethyl)-3-cyclohexyl-1-nitrosourea

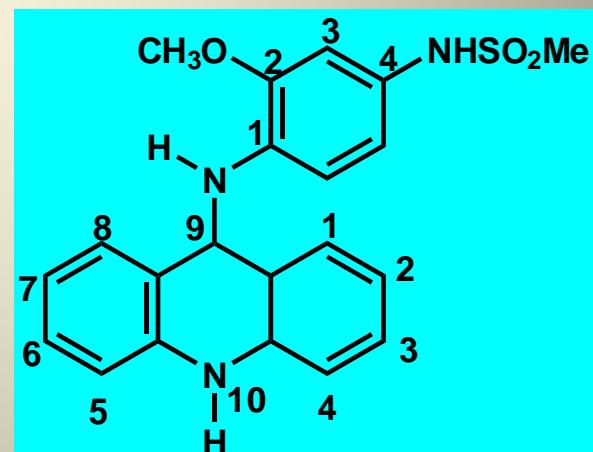
- Duration of Action:
 - BCNU: $t_{1/2} = \textcolor{red}{90 \text{ min.}}$
 - CCNU: $t_{1/2} = \textcolor{red}{16 \text{ hr.}}$
- Nitrosoureas possess high lipid solubility which allows BBB crossing, so they used mainly to treat **Brain tumors**

B. Groove Binding Agents (DNA Intercalators)

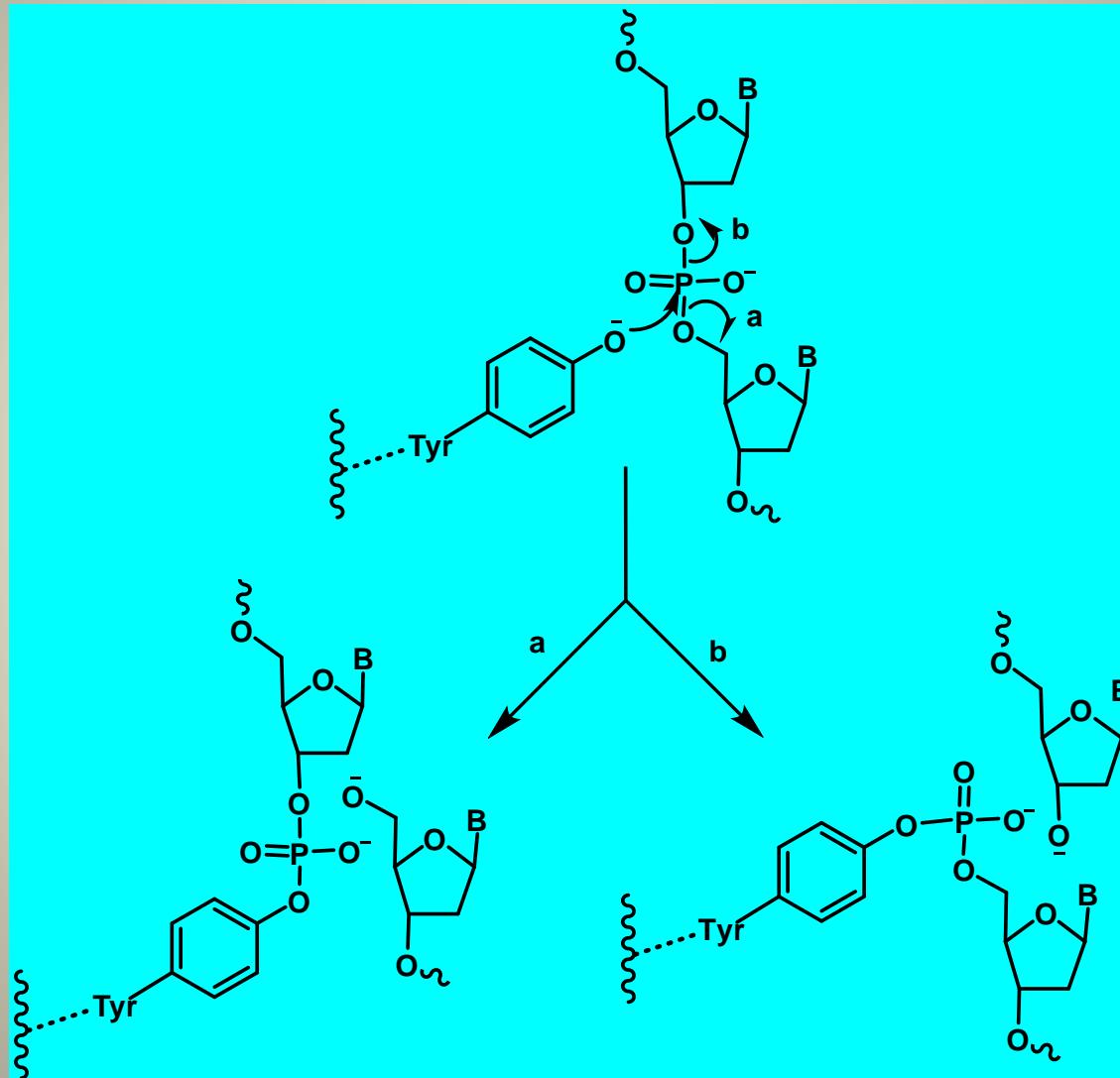
- Polycyclic planar compounds have the ability to insert or intercalate through the grooves of the base pairs of the DNA double helix.
- This insertion process will activate **Topoisomerase I & II** enzymatic system which catalyzes DNA strand cleavage.
- Acridines such as **Amsacrine** proved to possess this DNA intercalation activity.

Amsacrine

9-[(2-Methoxy-4-methylsulphonylamino) aniline]acridine.



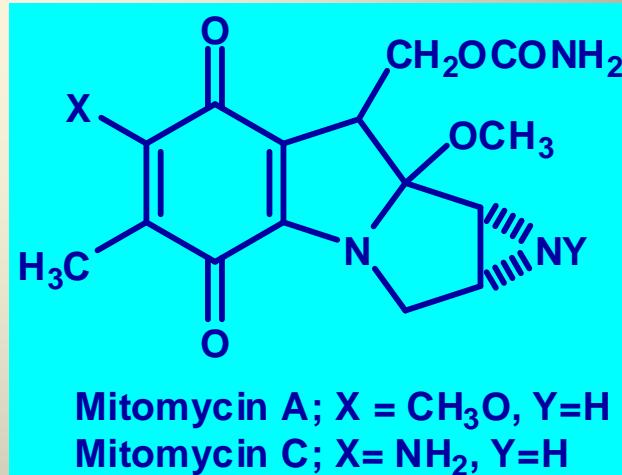
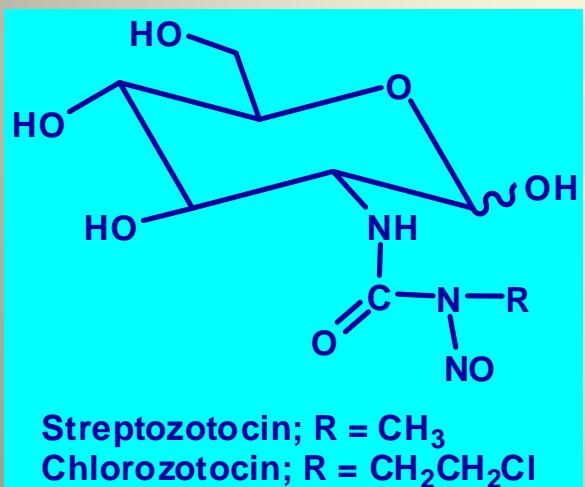
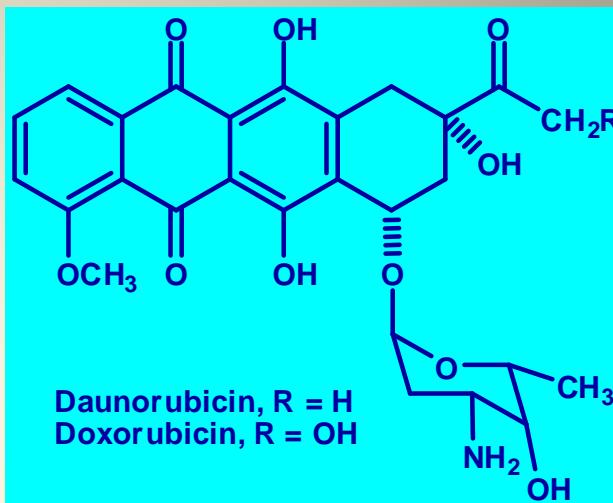
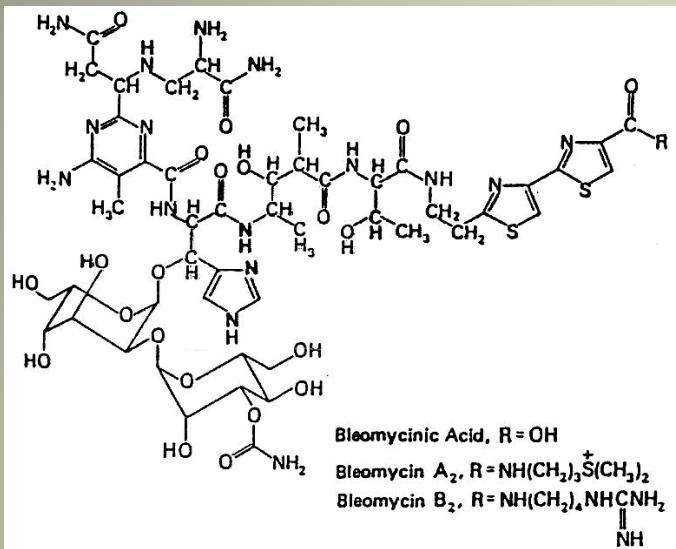
- The tyrosinyl moiety carried by Topoisomerases could catalyze such DNA cleavage.



C. DNA Strand Breakers

- A group of natural products are characterized by their ability to intercalate into the DNA and initiate a series of free radical destruction of the DNA strand **preventing the process of mitosis** and consequently preventing cell division.
- Those Natural products are:
 - ❖ Antibiotics
 - Bleomycins, Daunorubicins, Mitomycins, Streptozotocin
 - ❖ Vinca Alkaloids
 - Vinblastine, Vincristine

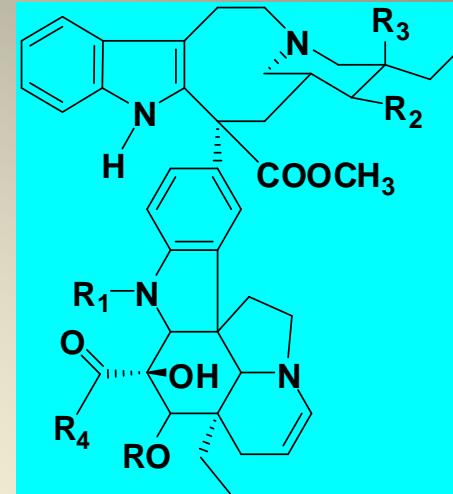
❖ Antibiotics



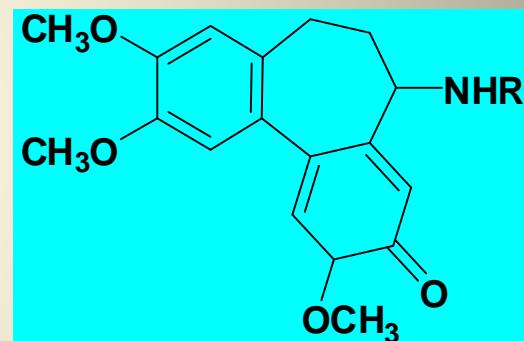
❖ Vinca Alkaloids and their Analogs

Vincrestin

Vinblastin

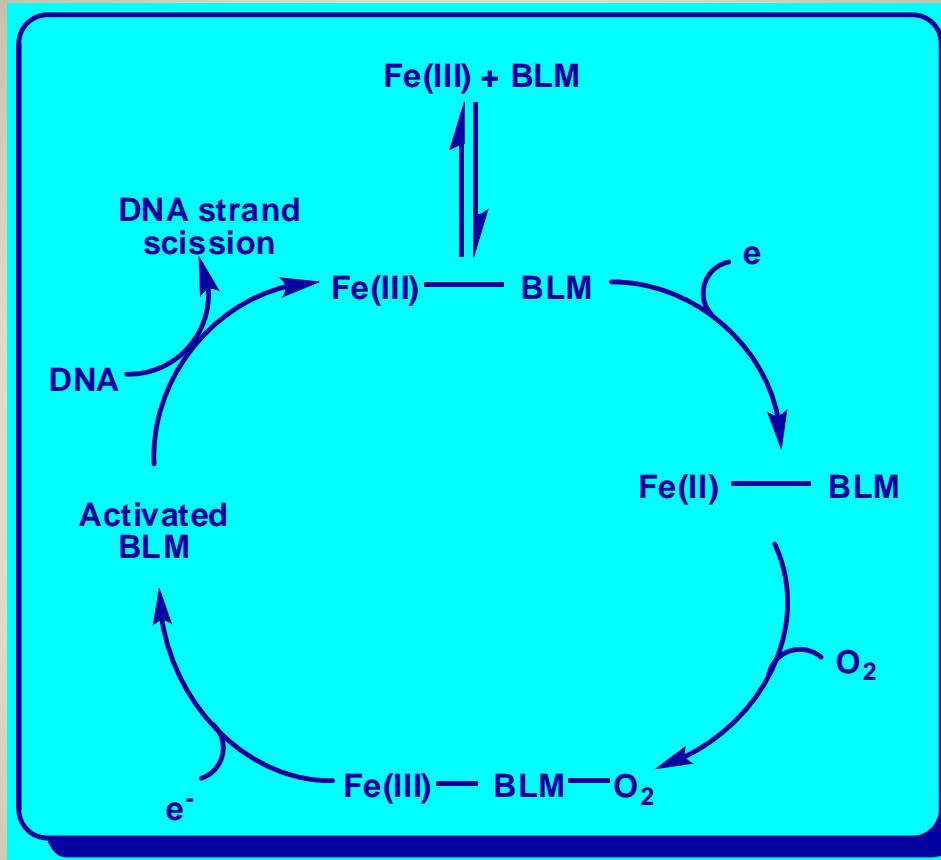


- Colchicine, obtained from **Colchicum autumnale**, Its main use is in terminating acute attacks of gout.
- Colchicines have an unusual tricyclic structure containing a **tropolone ring**.
- Colchicines and its derivative demecolcine (colcemid) is active against myelocytic leukemia.
- They inhibit mitosis at metaphase by disorienting the organization of the **mytotic spindle**.



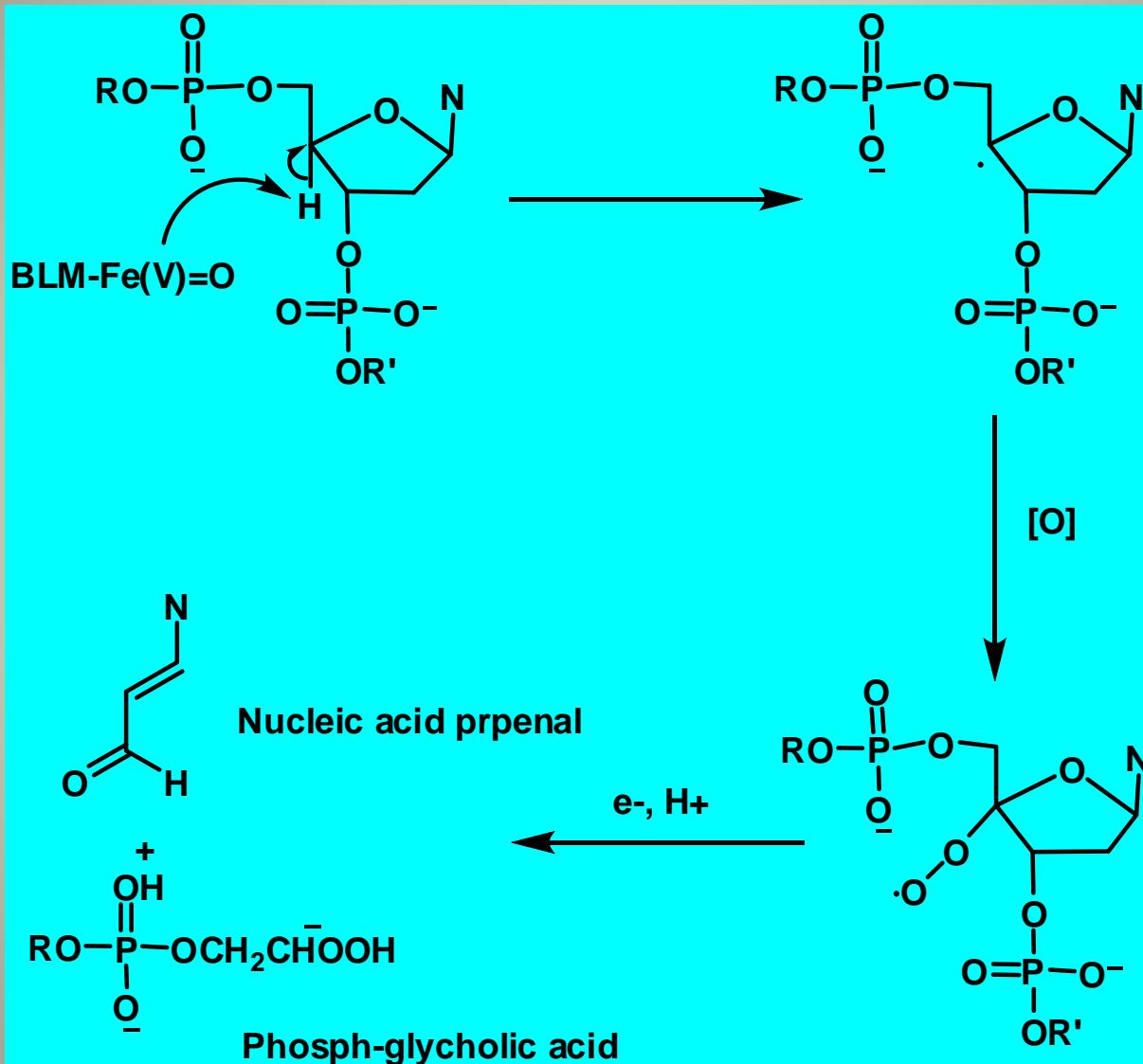
Colchicine, R = COCH₃,
Colcemid, R = CH₃

Mode of Action of Strand Breakers



Cycle of events involved in DNA cleavage by bleomycin (BLM)

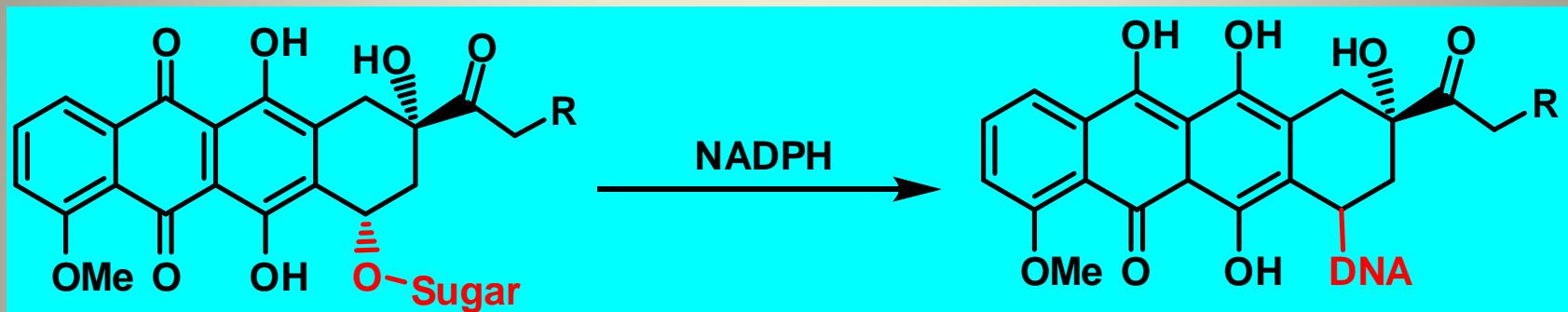
DNA strand Scission by Activated Bleomycin



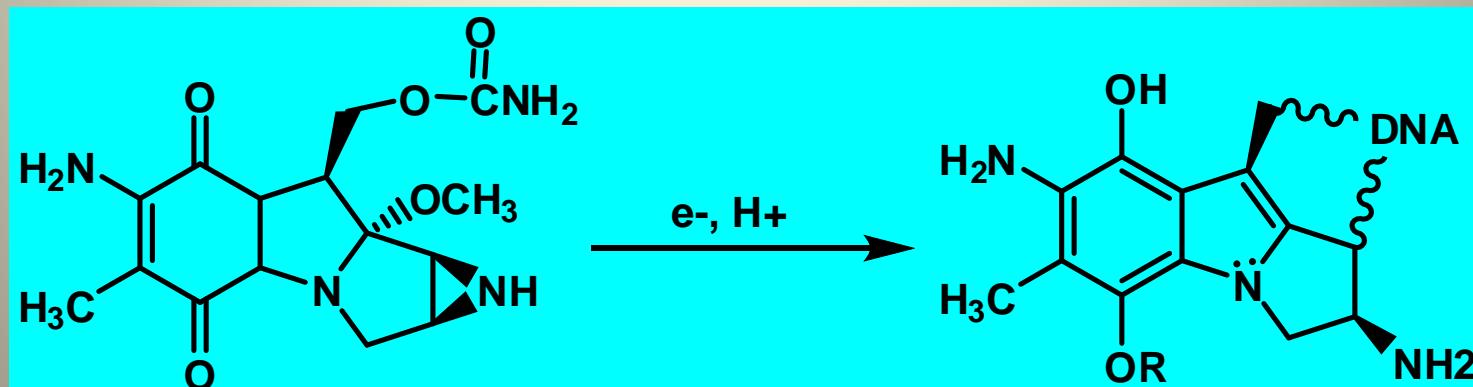
Mode of Action of Antimitotic Agents

i. Daunorubicin Analogs

Anthracycline antitumor agents as bioreductive alkylators



ii. Mitomycin analogs

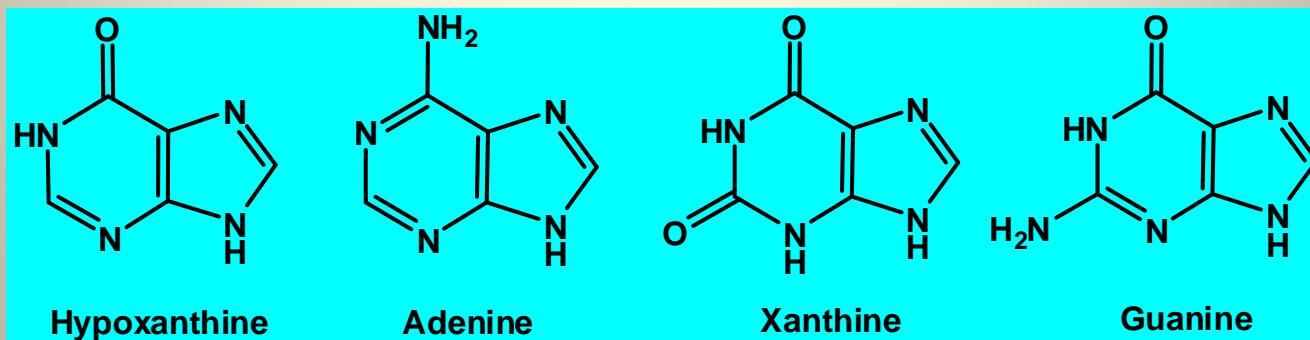


II. Antimetabolites

- Those are class of compounds which are structurally related to natural occurring substances found in normal cells.
- Antimetabolites compete with those natural cell components for the active sites on enzyme(s) or receptor(s), and they might incorporate into the nucleic acids to disrupt their cellular functions.
- Antimetabolites could be classified into:
 - A. Purine Antagonists
 - B. Pyrimidine Antagonists
 - C. Folic acid Antagonists

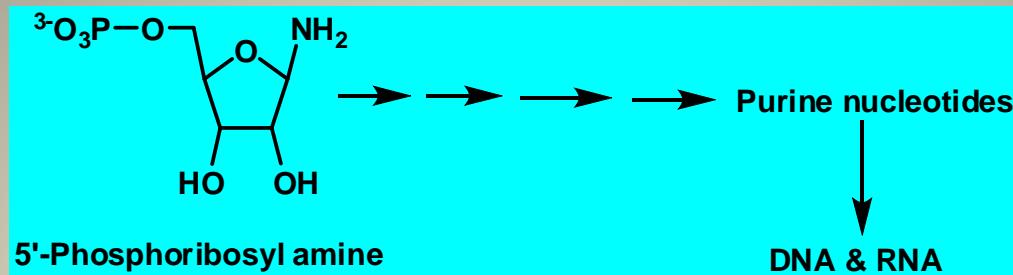
A. Purine Antagonists

- Those are group of compounds which are structurally related to the natural purine bases (**hypoxanthine**, **adenine**, **xanthine** and **guanine**) and compete with their cellular functions.
- They are proved to inhibit **aminotransferase**, **adenylsuccinate synthase**, **adenylsuccinate lyase** and **inosine monophosphate dehydrogenase** enzymatic systems leading to protein synthesis inhibition followed by cell death.

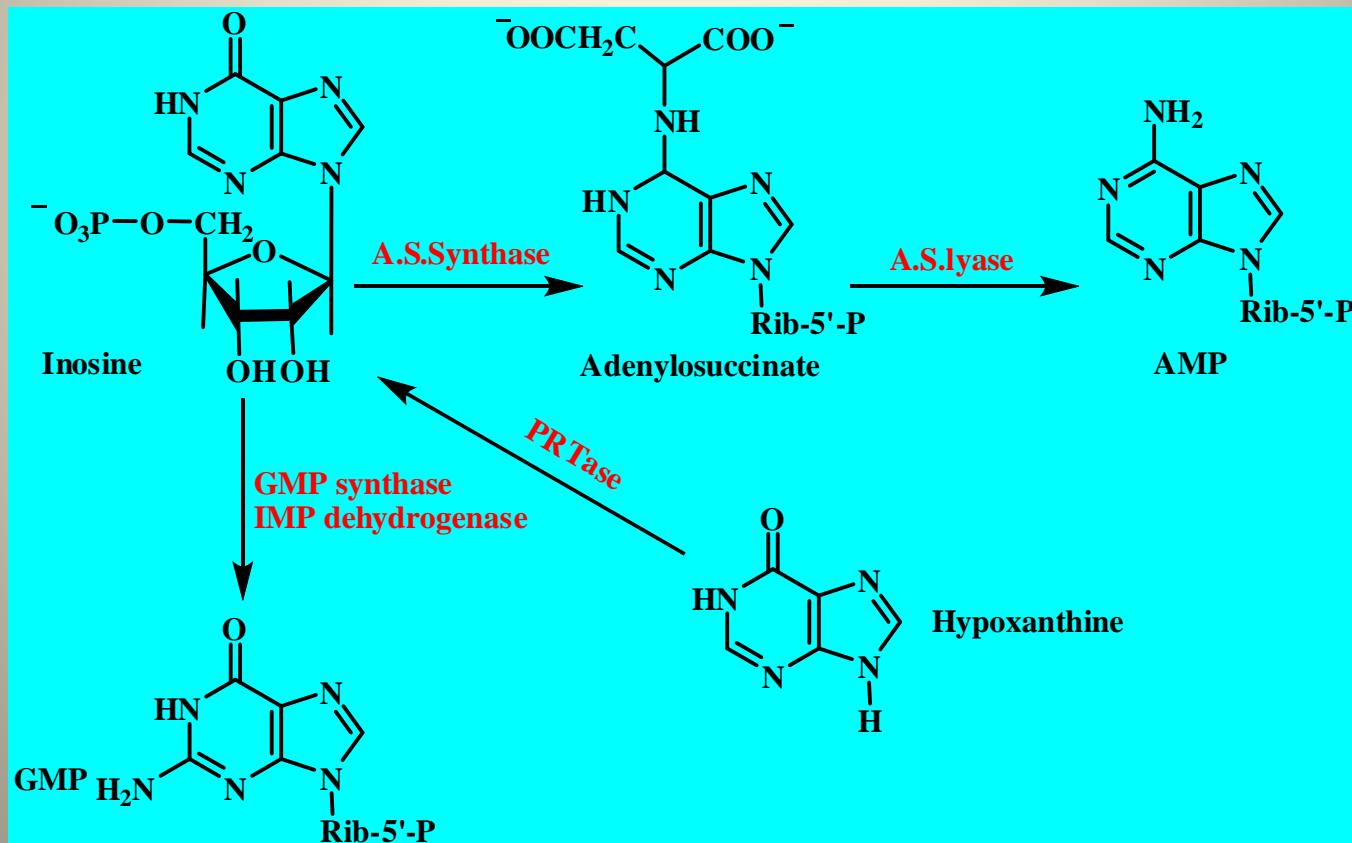


- Normal cell can get its need from these bases through:
 - De Novo Purine Synthesis
 - Salvage Purine Synthesis

De Novo Purine Synthesis

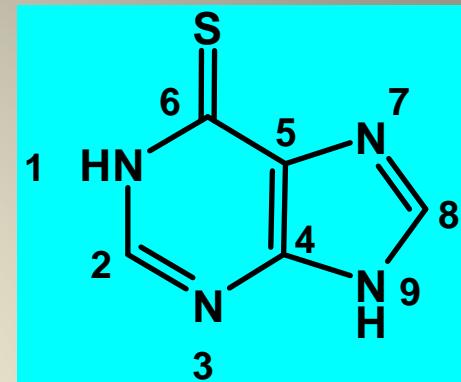


Salvage Purine Synthesis

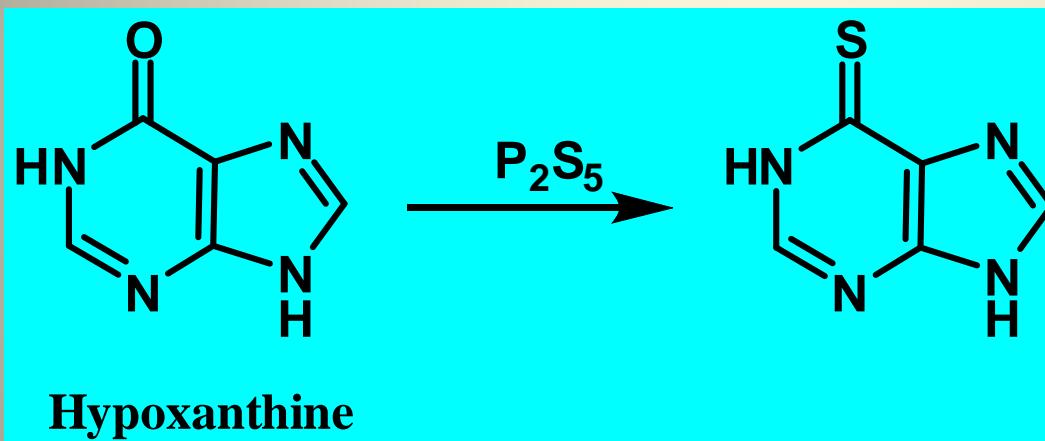


1. Mercaptopurine, Lukerine

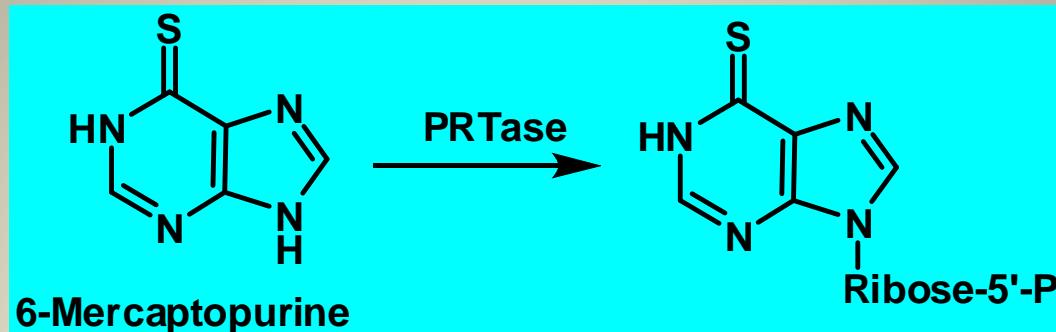
6-Mercaptopurine or purine-6-thiol.



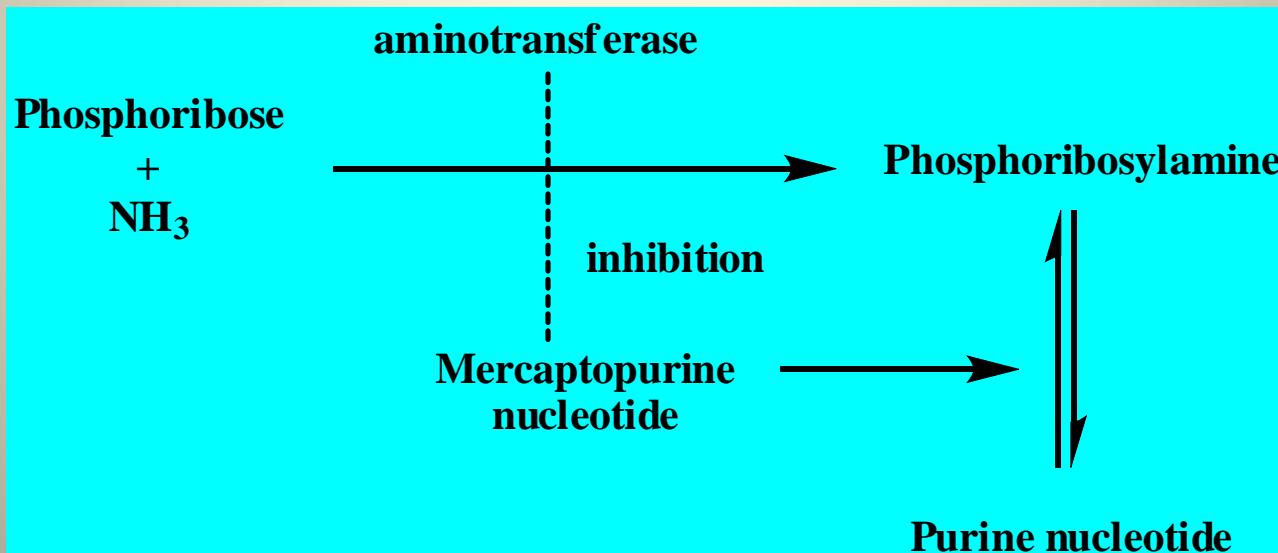
Synthesis



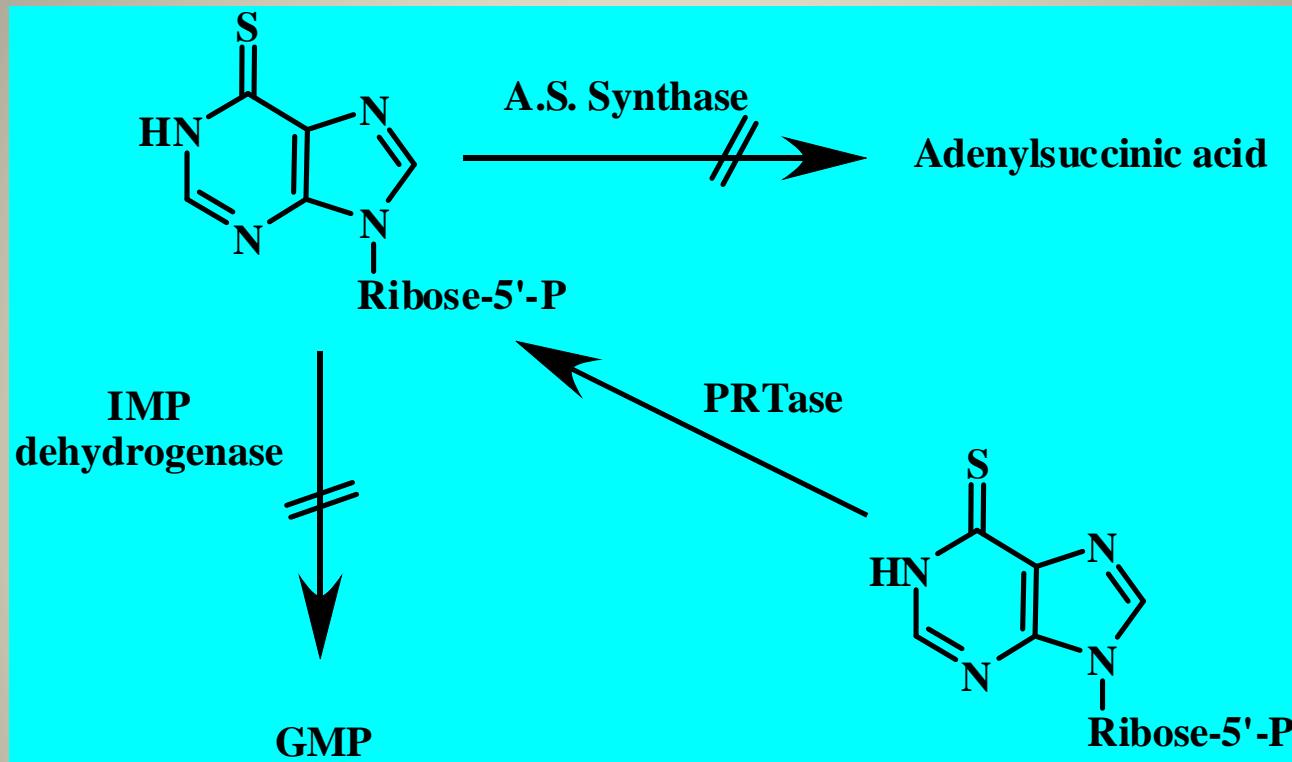
Mechanism of Action



1. Inhibition of De Novo Synthesis

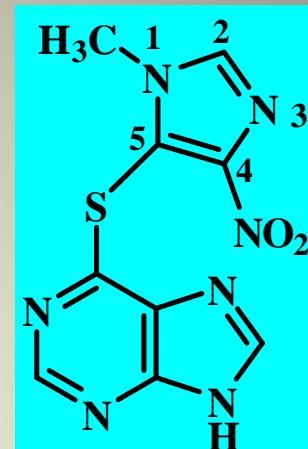


2. Inhibition of Salvage Purine Synthesis

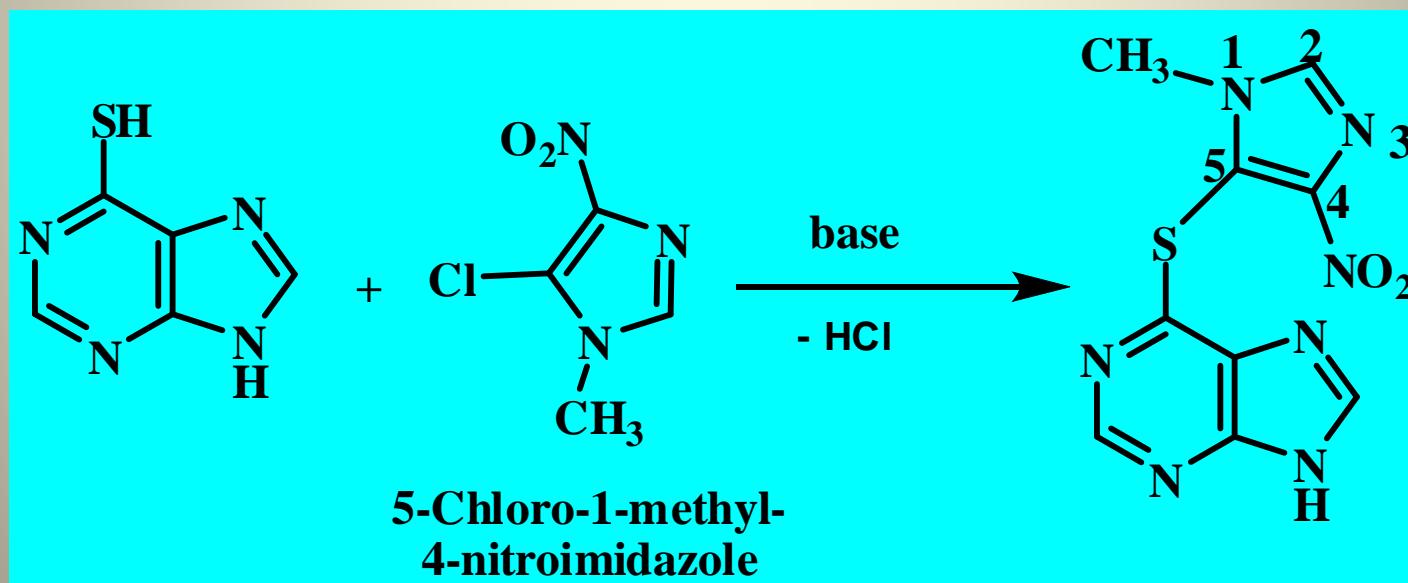


2. Azathioprine, Imuran

6-[(1-Methyl-4-nitroimidazol-5-yl)thio]purine

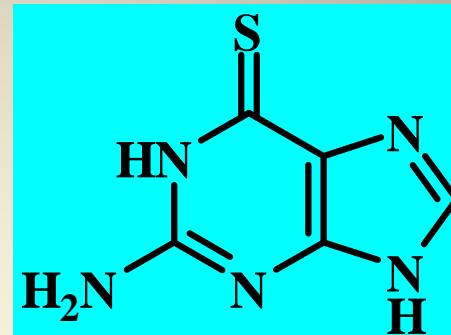


Synthesis



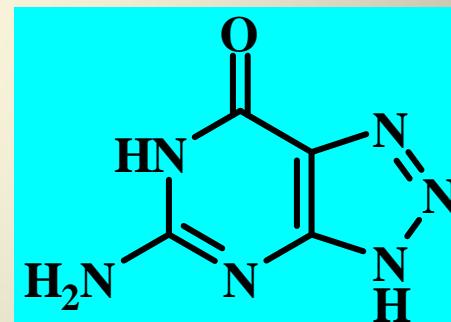
3. Thioguanine

2-Amino-6-mercaptopurine



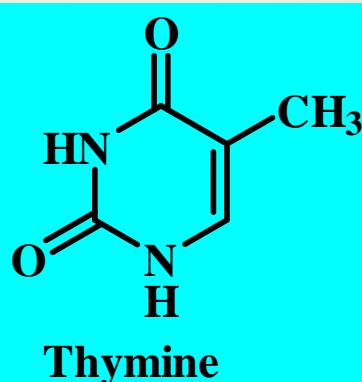
4. Azaguanine, Guanazole

2-Amino-6-hydroxy-8-azapurine

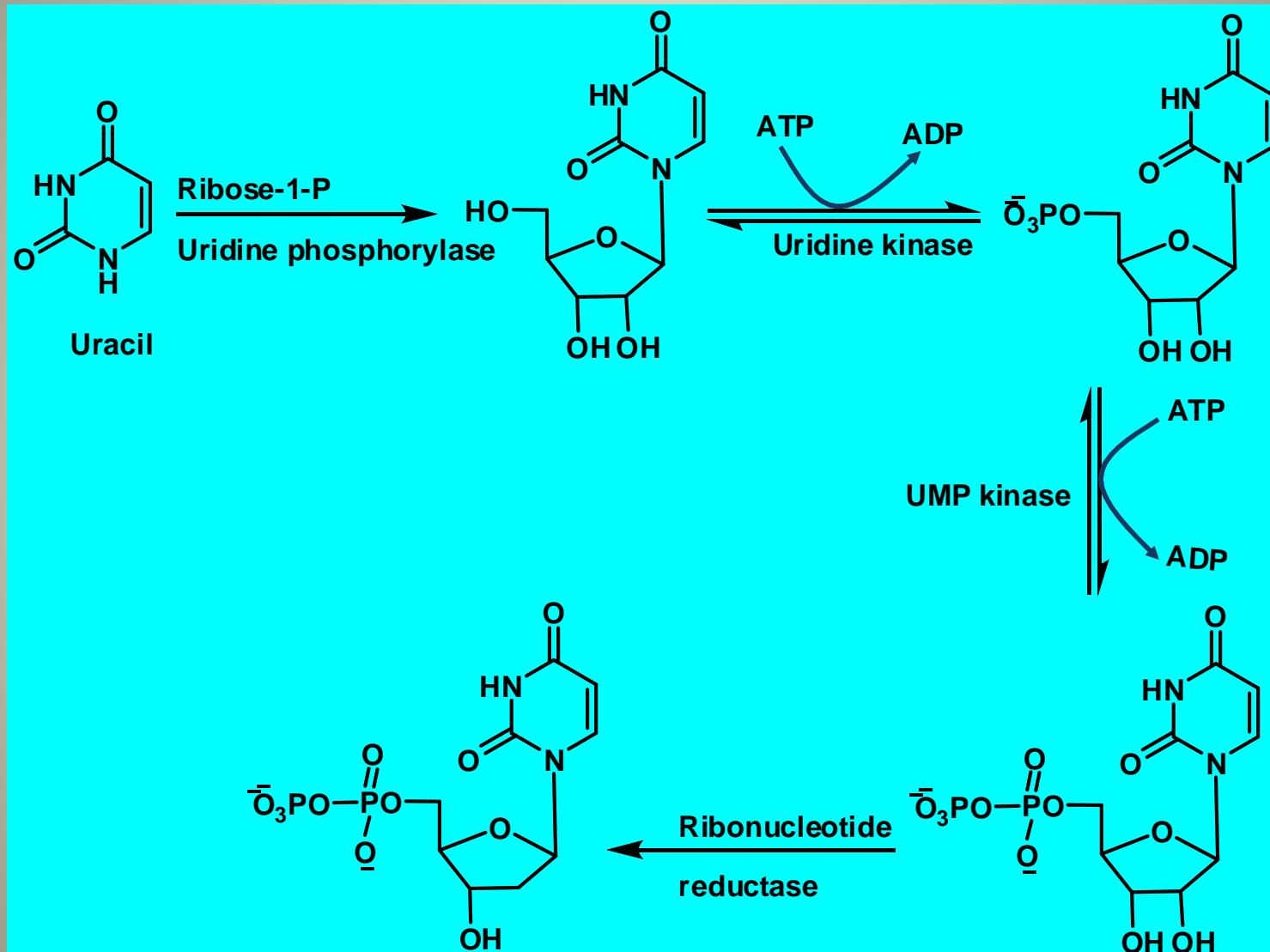


B. Pyrimidine Antagonists

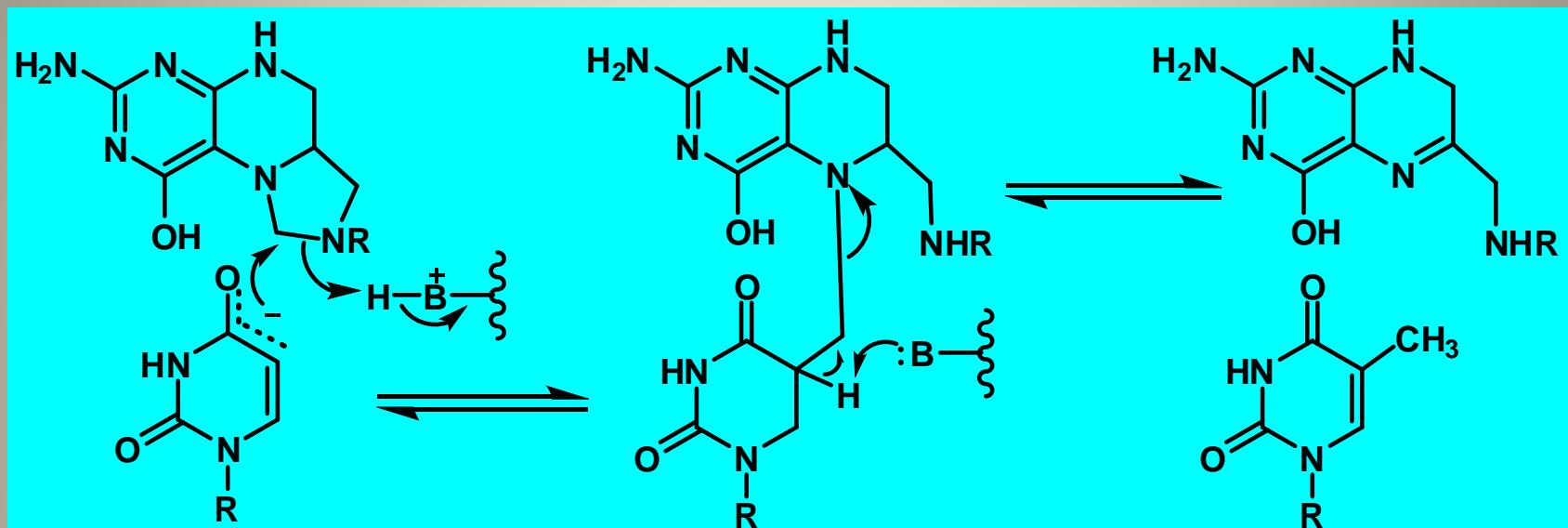
- They resemble the natural pyrimidine (**uracil**, **thymine** and **cytosine**) in structures and compete with their cellular functions leading to the inhibition of two vital enzymatic systems responsible for the production of thymine from uracil.
- These two enzymes are **Ribonucleotide Reductase** and **Thymidylate Synthase**.
 - The metabolic role of ribonucleotide reductase in the production of deoxysugars.
 - The metabolic role of thymidylate synthase in the production of thymidine from uridine.



The metabolic role of **Ribonucleotide Reductase** enzyme

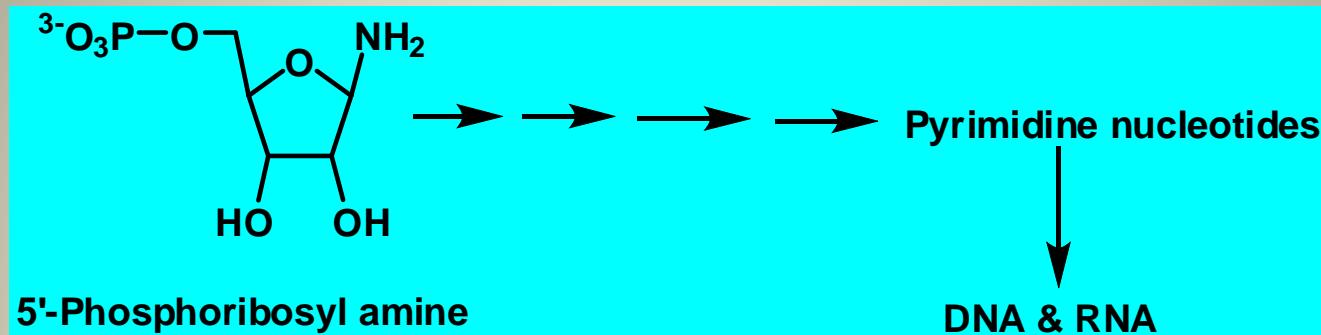


The metabolic role of Thymidylate Synthase enzyme

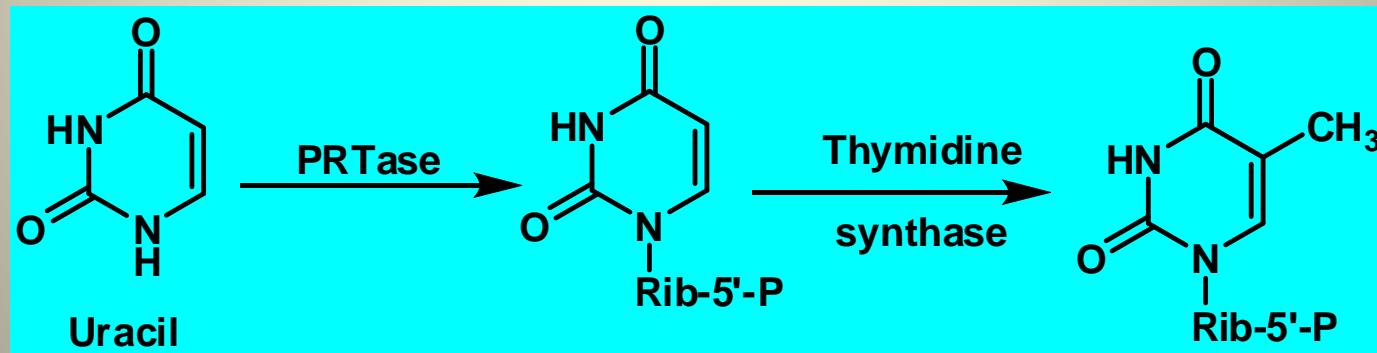


- Normal cells can get its need from these bases through:

De Novo Pyrimidine Synthesis



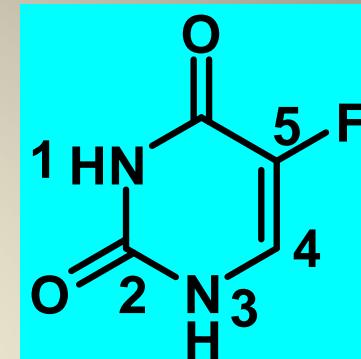
Salvage Pyrimidine Synthesis



1. Fluorouracil, Fluropex

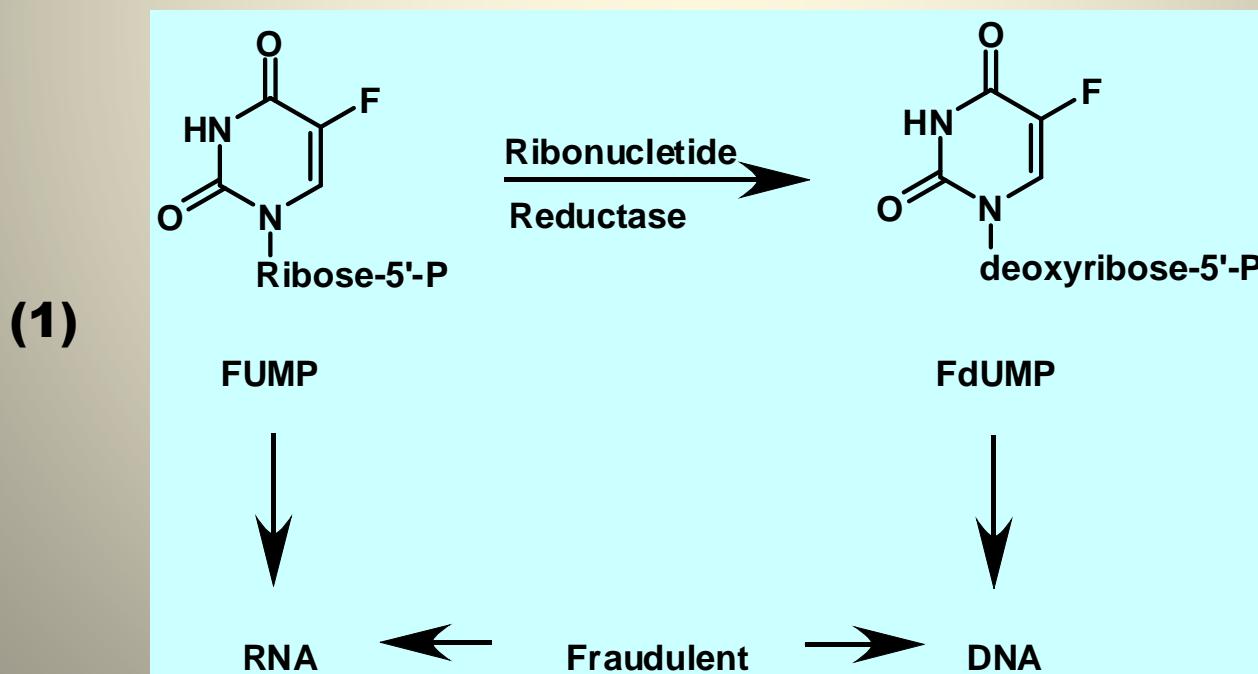
5-Fluorouracil

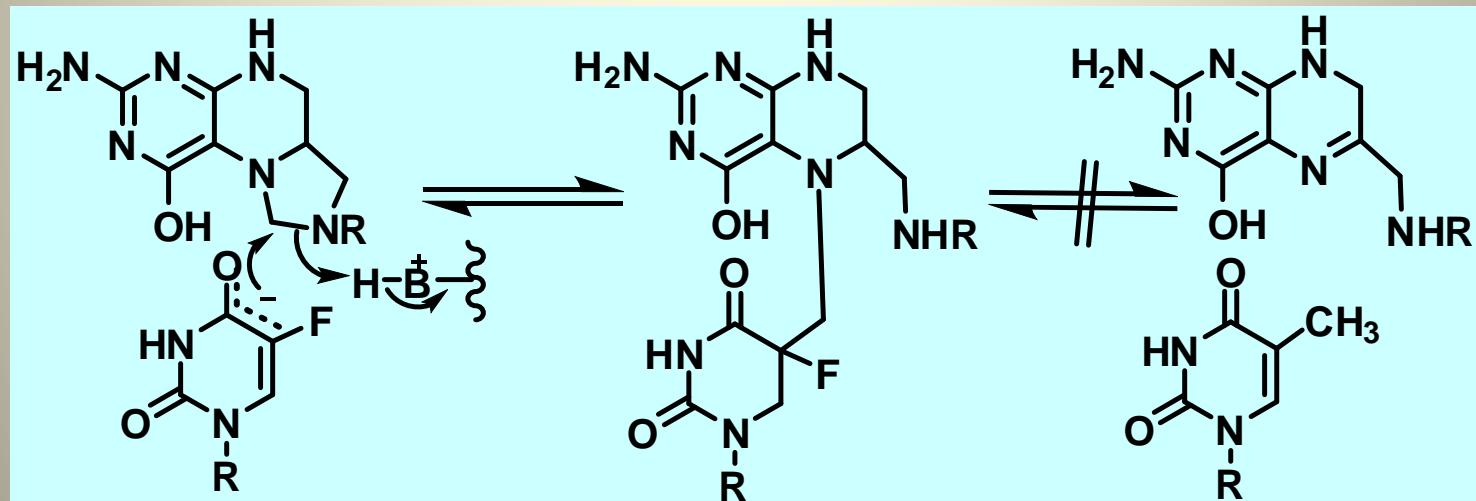
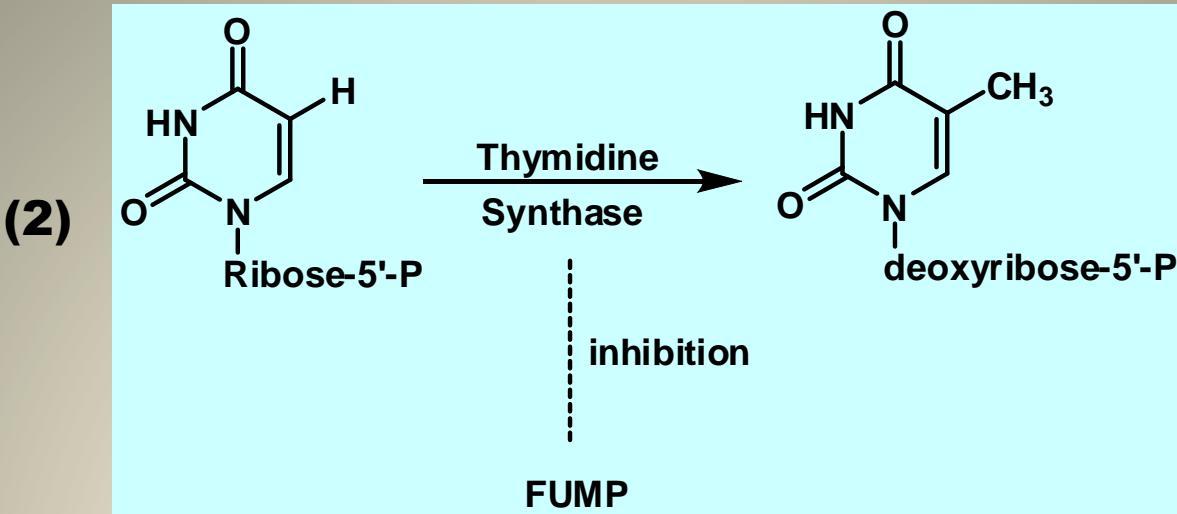
5-Fluoro-2,4 (*1H, 3H*)-pyrimidindione.



Mode of action

5-Fluorouracil is multimodal, exerting its effect by more than one mode

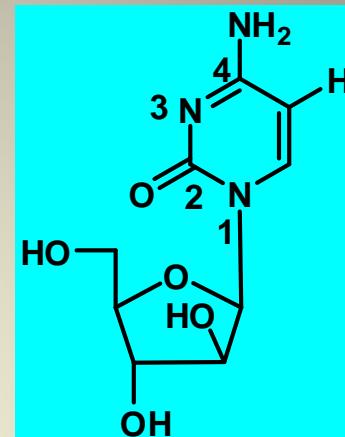




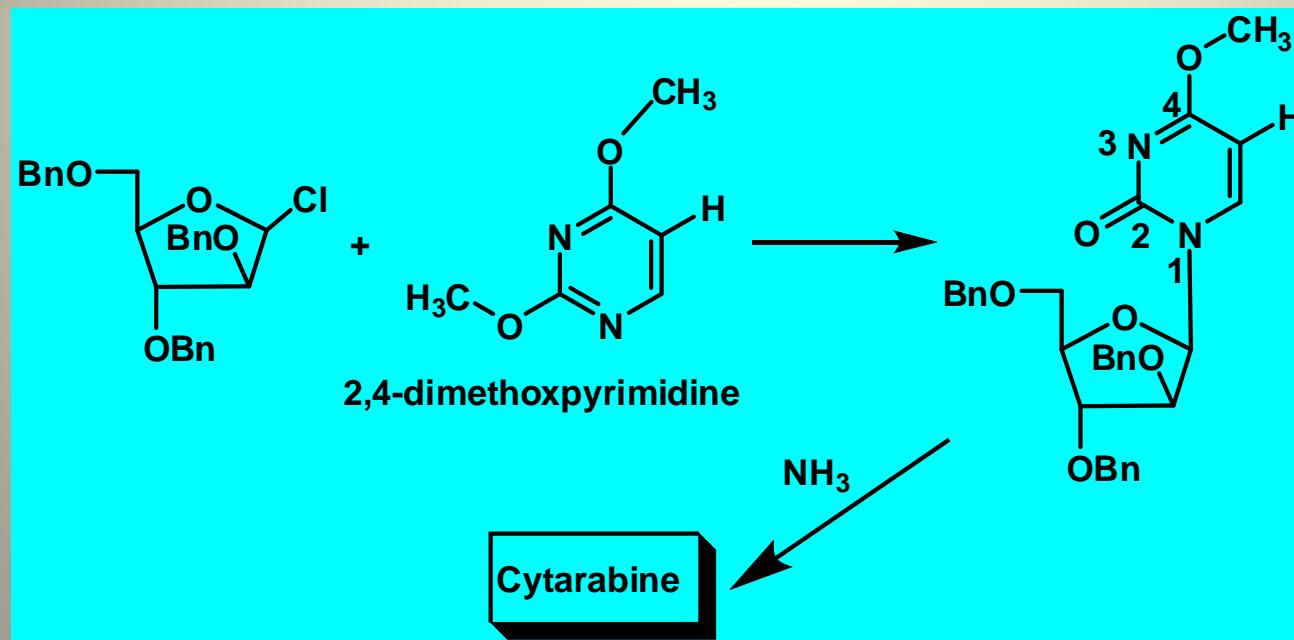
Proposed mechanism for the inactivation of thymidylate synthase by 5-fluoro-2'-deoxyuridylate.

2. Cytarabine, Cytosar

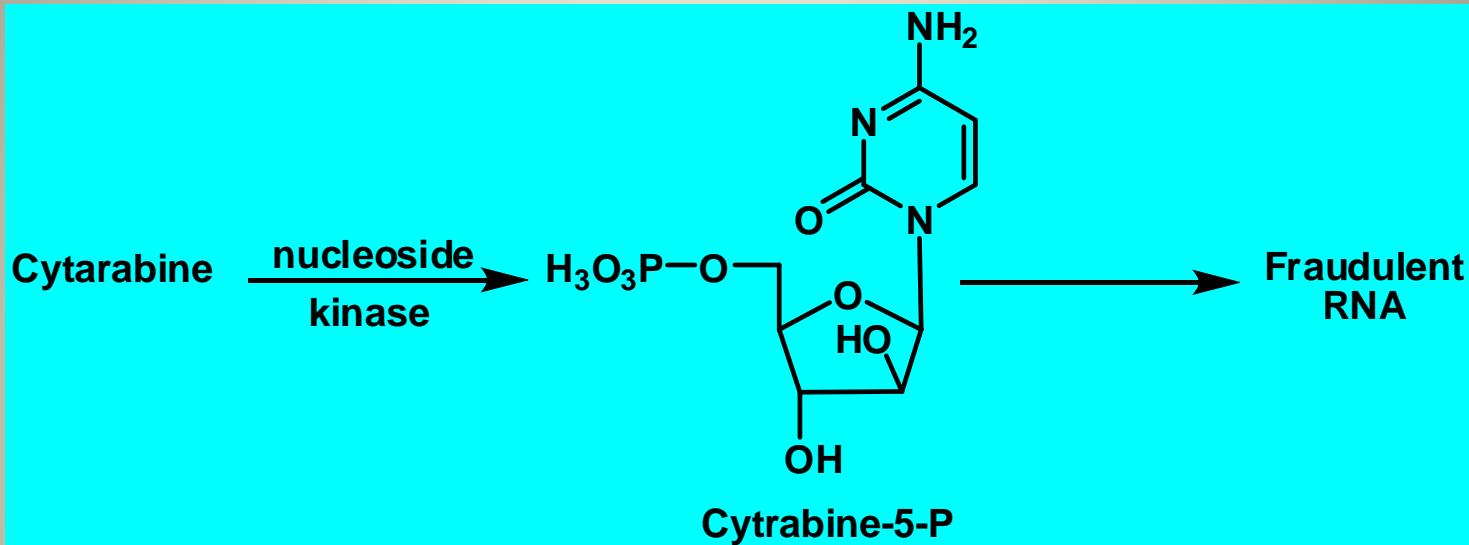
4-Amino-1-*B*-D-arabinofuranosyl-2(1*H*)-pyrimidinone



Synthesis



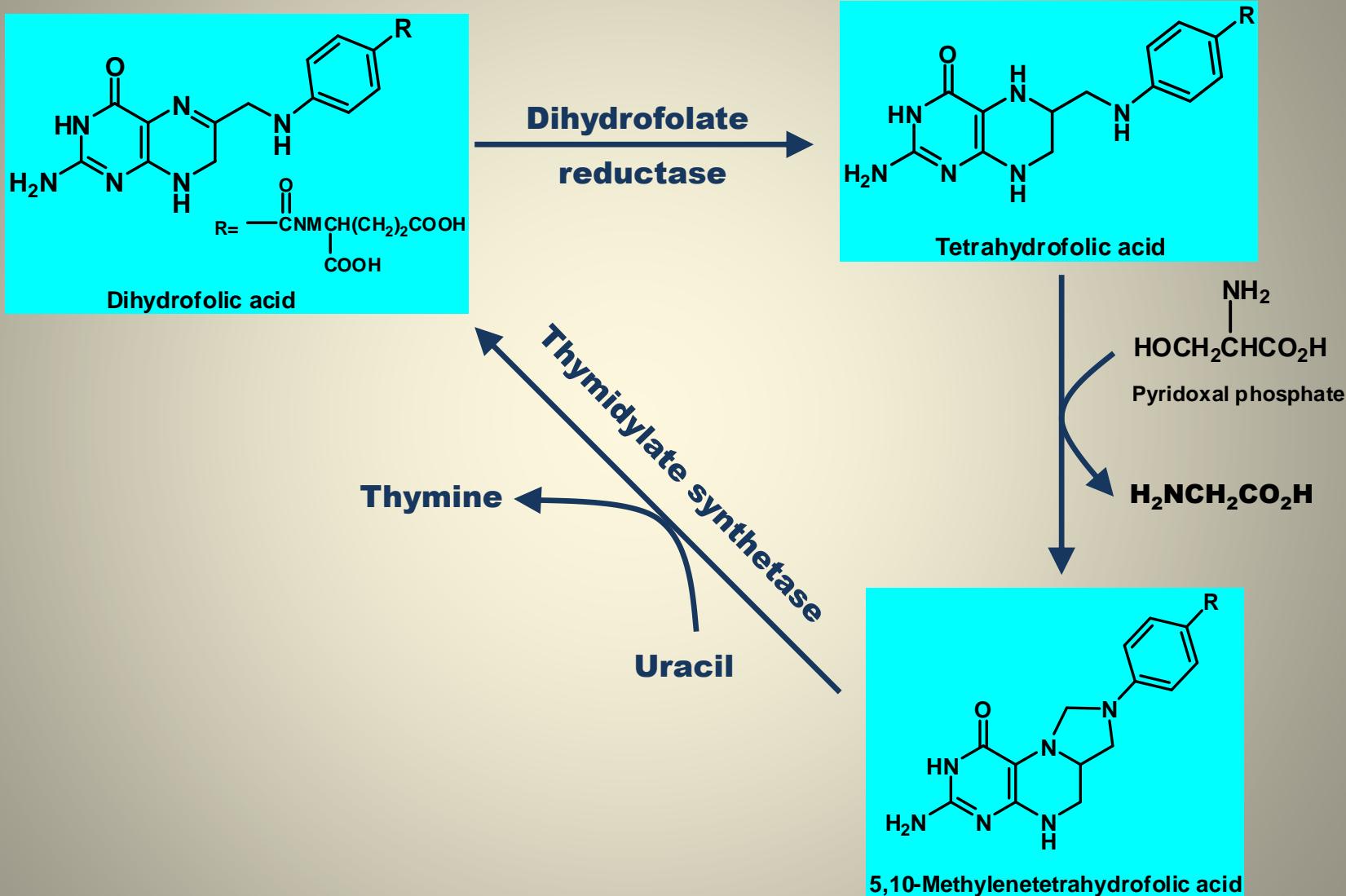
Mode of Action



C. Folic Acid Antagonists

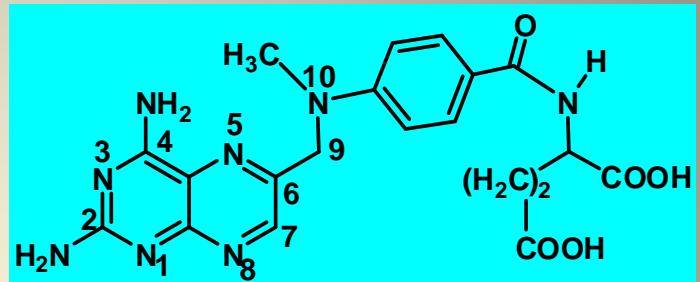
- Folic acid metabolism is an important source for **one carbon moiety** needed to convert uracil into thymine.
- Inhibition of folic acid metabolism, (inhibition of **dihydrofolate reductase**) deplete the biological systems from thymine –a pyrimidine base- very much needed for nucleic acid biosynthesis.
- Folic acid antagonists could be classified into:
 - A. Dihydrofolate reductase inhibitors
Methotrexate
 - B. Thymidylate synthetase inhibitors
5-Fluorouracil, Fluoroplex

The metabolic role of Dihydrofoalte Reductase enzyme

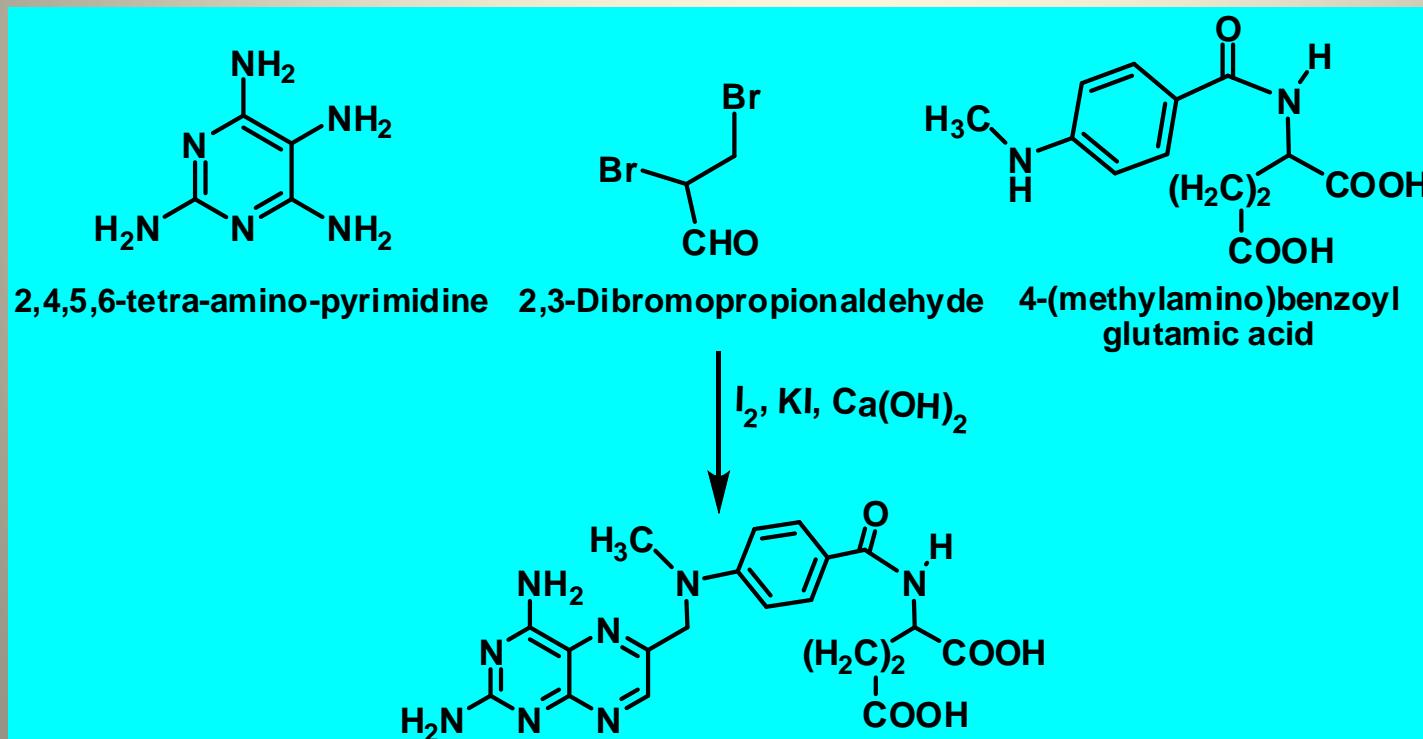


1. Methotrexate, Mexate, Amethopterin

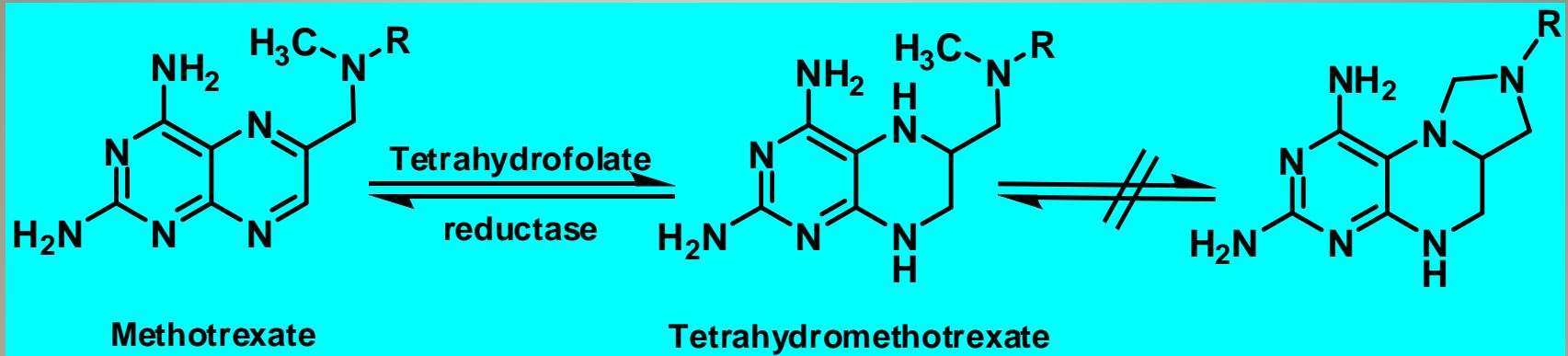
4-Amino-10-methylpteroylglutamic acid



Synthesis



Mode of Action



Contraindications:

Salicylates and sulfonamides increase methotrexate toxicity by:

1. Inhibiting its renal tubular secretion.
2. They displace methotrexate from plasma protein binding.

2. Fluorouracil, Fluoroplex

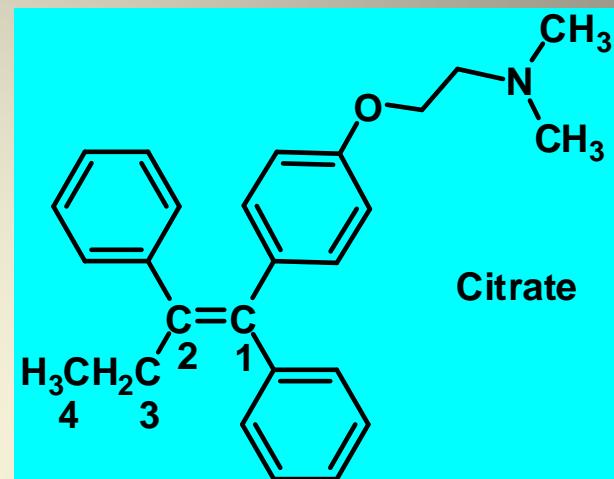
As discussed before

III. Hormones

- Some cancers are responsive to sex hormone treatment.
- Hormones control the dissemination of cancer but suffer from side effects as:
 1. Androgens → masculinizing effect
 2. Estrogens → feminizing effect
 3. Adrenocorticoids → salt and water retention
- Tamoxifen citrate is a nonsteroidal drug with anti-estrogenic effect used as anticancer

Tamoxifen citrate, Nolvadex

2-[4-(1,2-Diphenyl-1-butenyl)phenoxy]-N,N-dimethylethanamine citrate.



Mode of Action

- Tamoxifen is an estrogen receptor antagonist, blocks the growth promoting effects of estrogen on tumors.
- Uses: Advanced breast carcinoma.

Summary of Anticancer Drug Mechanisms and Sites

